

Ring currents in *periodo*-hydrocarbons

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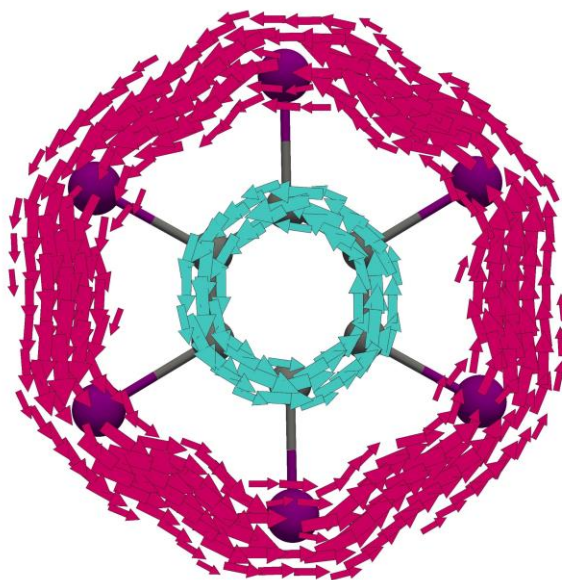
Introduction

- Double aromaticity – aromaticity composed of two circularly delocalized orbitals.[‡]

[‡]J. Chandrasekhar, E. D. Jemmis. P. v. R. Schleyer, *Tetrahedron Letters*, 1979, 20, 39, 3707-3710

Introduction

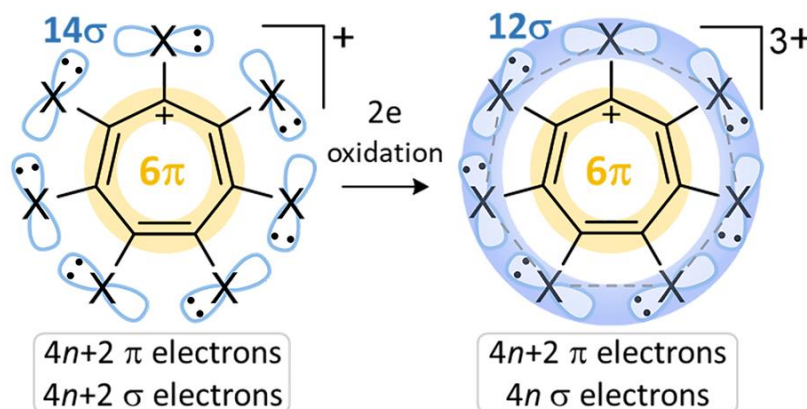
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- Double aromaticity in halogen substituted benzene dications derivatives has been extensively investigated.*



* I. Ciofini, P. P. Lainé and C. Adamo, *Chem Phys Lett*, 2007, 435, 171–175;
R. W. A. Havenith, P. W. Fowler, S. Fias and P. Bultinck, *Tetrahedron Lett*, 2008, 49, 1421–1424;
M. Orozco-Ic, J. Barroso, R. Islas and G. Merino, *ChemistryOpen*, 2020, 9, 657–661.

Introduction

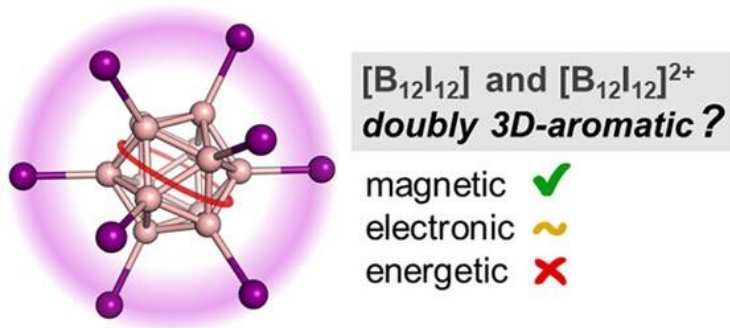
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- Double (anti)aromatic character is later investigated in halogen substituents of the tropylium cations and the cyclooctatetraene cations.[#]



[#] S. Escayola, N. Proos Vedin, A. Poater, H. Ottosson and M. Solà, *J Phys Org Chem*, 2023, 36, e4447.
P. W. Fowler and R. W. A. Havenith, *J Phys Chem A*, 2021, 125, 6374–6383.

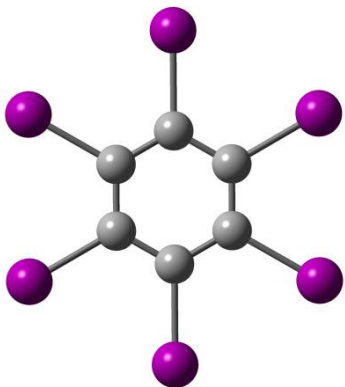
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- Double aromaticity in halogen substituted benzene dications derivatives has been extensively investigated.*
- Double (anti)aromatic character is later investigated in halogen substituents of the tropylium cations and the cyclooctatetraene cations.[#]
- Possibility of double 3D-aromaticity is analysed in $[\text{B}_{12}\text{I}_{12}]$ and $[\text{B}_{12}\text{I}_{12}]^{2+}$.[§]

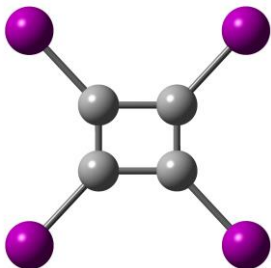


[§] J. Poater, S. Escayola, A. Poater, F. Teixidor, H. Ottosson, C. Viñas and M. Solà, *J Am Chem Soc*, 2023, 145, 22527–22538.

Studied molecules

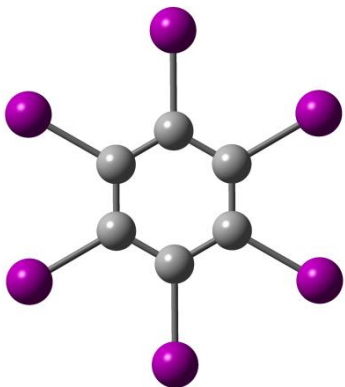


C₆I₆

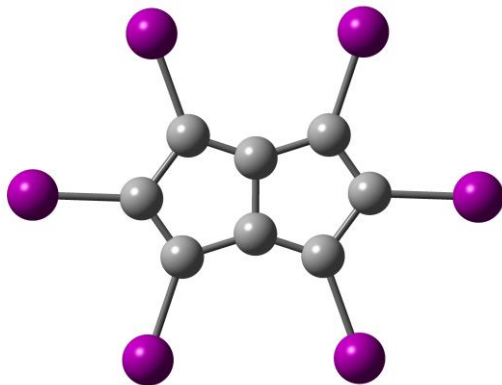


C₄I₄

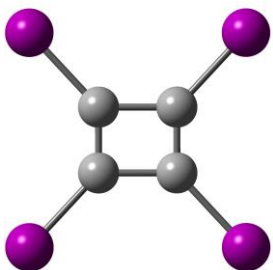
Studied molecules



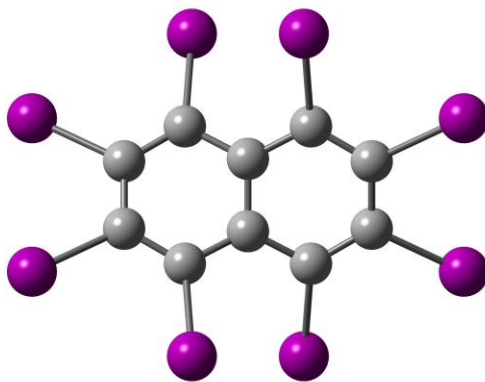
C₆I₆



C₈I₆

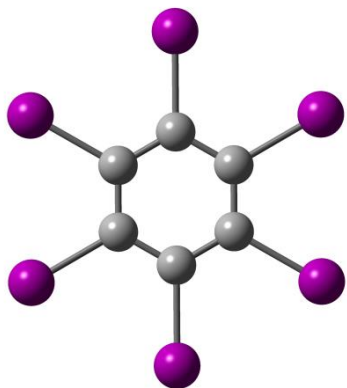


C₄I₄

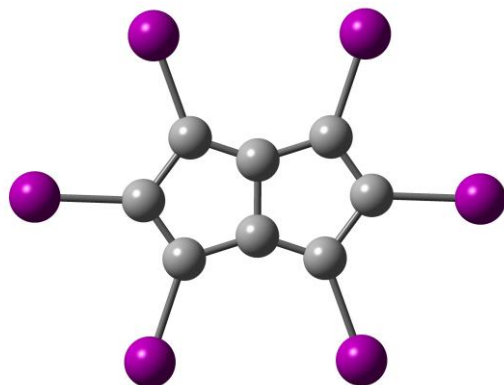


C₁₀I₈

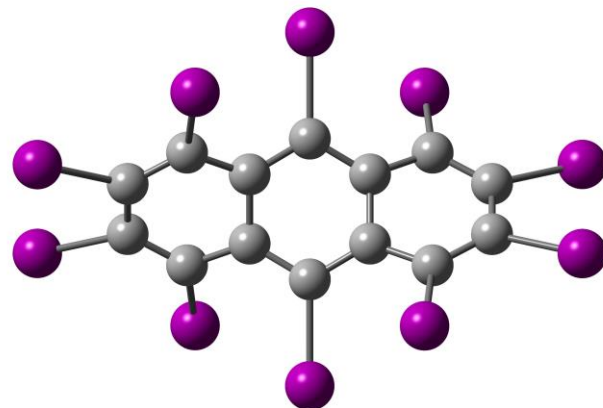
Studied molecules



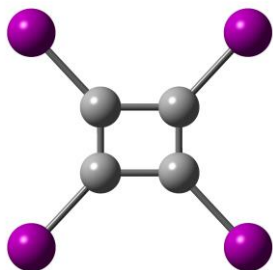
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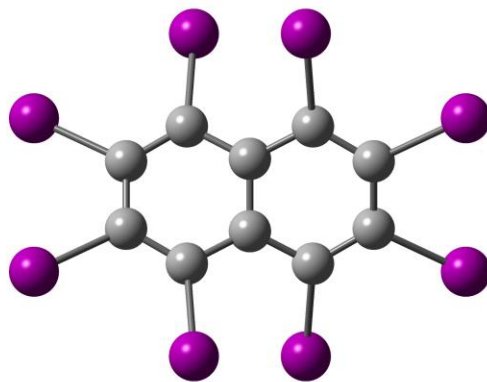
C₈I₆



C₁₄I₁₀

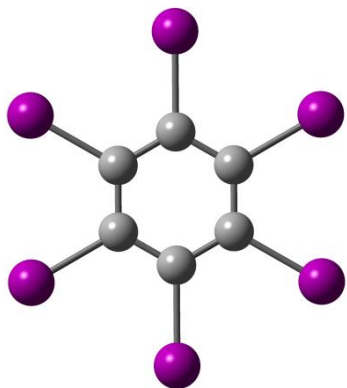


C₄I₄

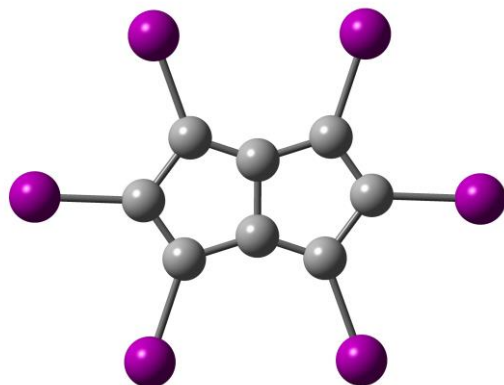


C₁₀I₈

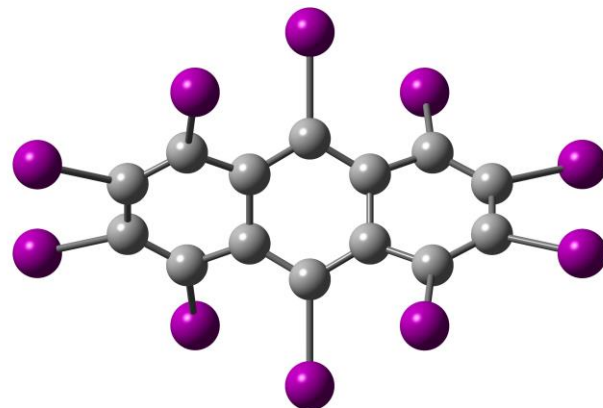
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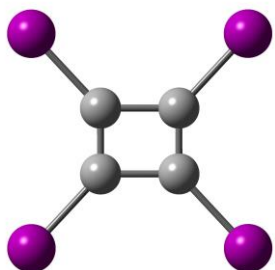
C₆I₆



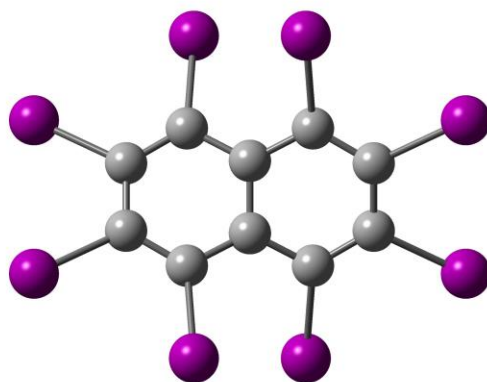
C₈I₆



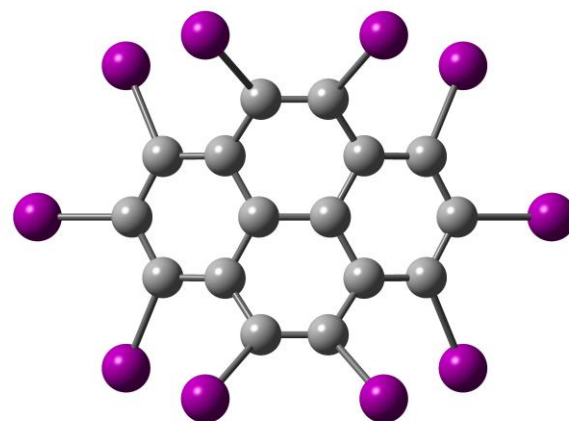
C₁₄I₁₀



C₄I₄



C₁₀I₈



C₁₆I₁₀

Computational details

- The geometries of studied molecules were optimized at B3LYP/def2TZVP level of theory using Gaussian09 program.

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 diatropic currents
 paratropic currents

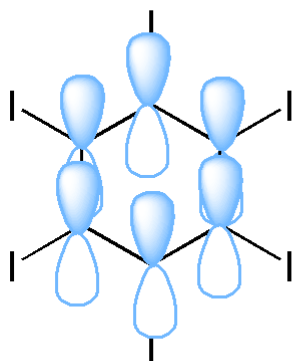
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-  diatropic currents
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- EDDB indices were computed using available software.

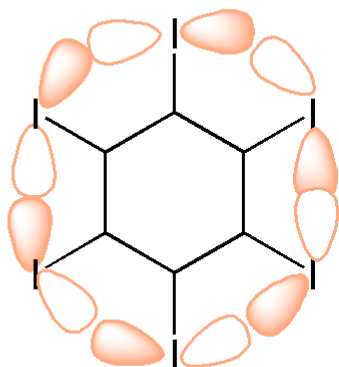
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- Calculations of I_{ring} indices were performed using in-house FORTRAN routine.

Periodo-benzene – σ and π notations

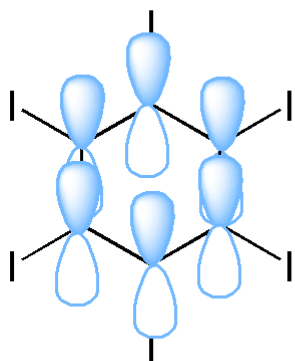


π orbitals

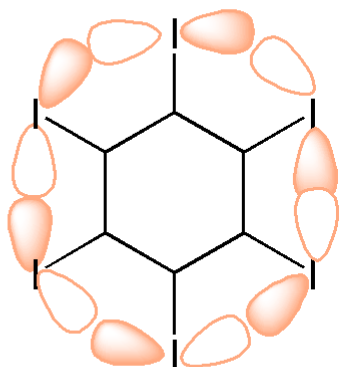


σ orbitals

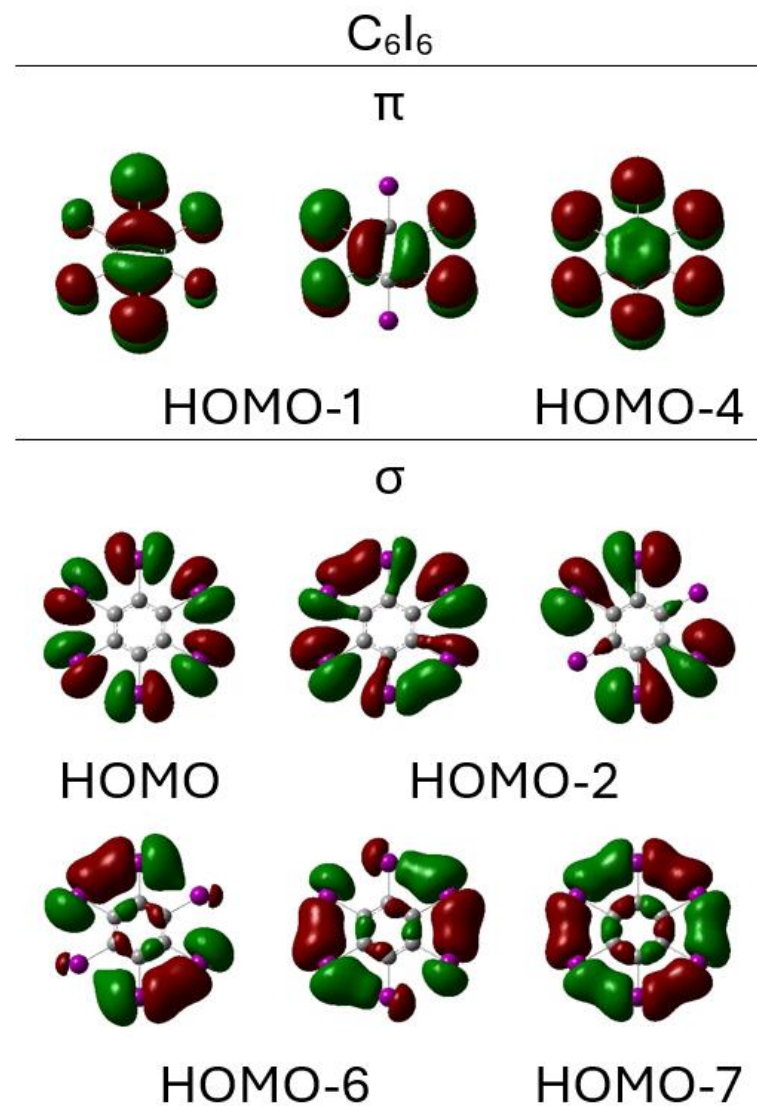
Periodo-benzene – σ and π notations



π orbitals



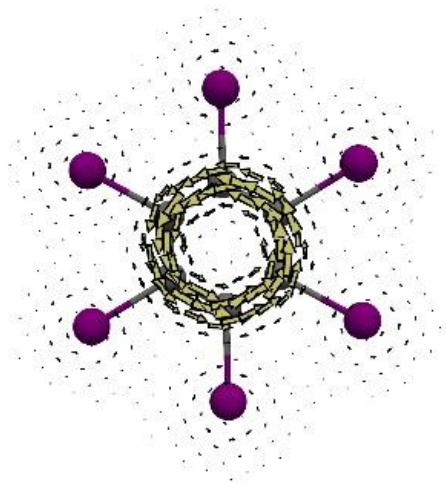
σ orbitals



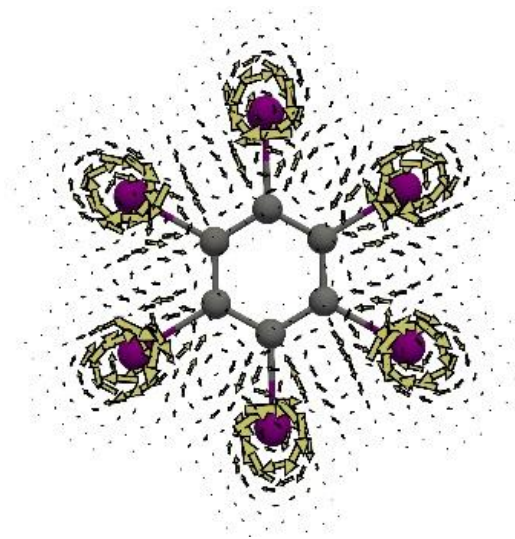
Periodo-benzene

π -currents

C_6I_6



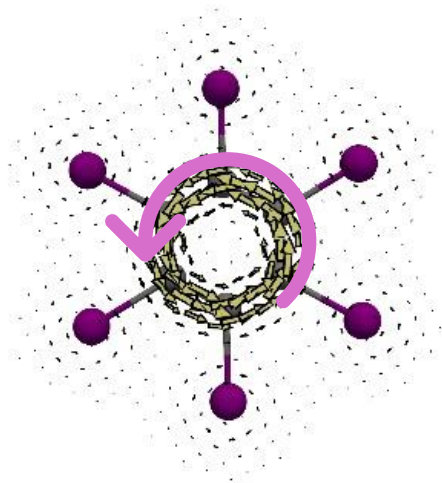
σ -currents



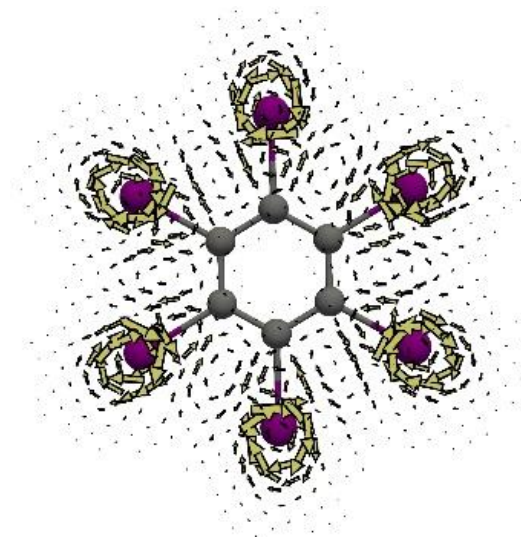
Periodo-benzene

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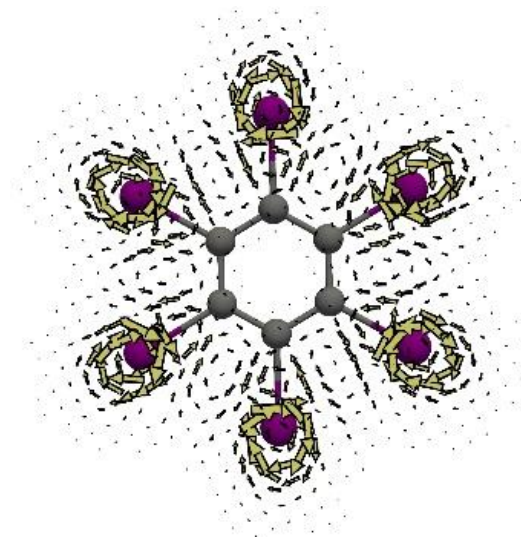
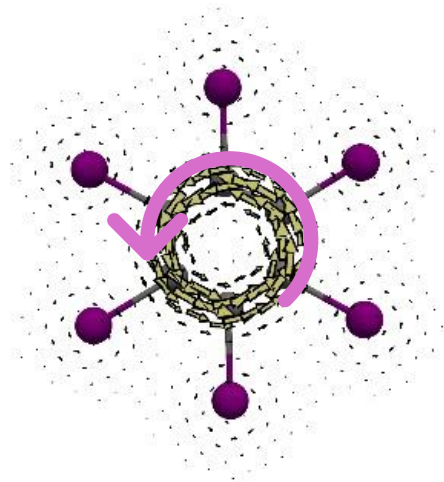


Periodo-benzene

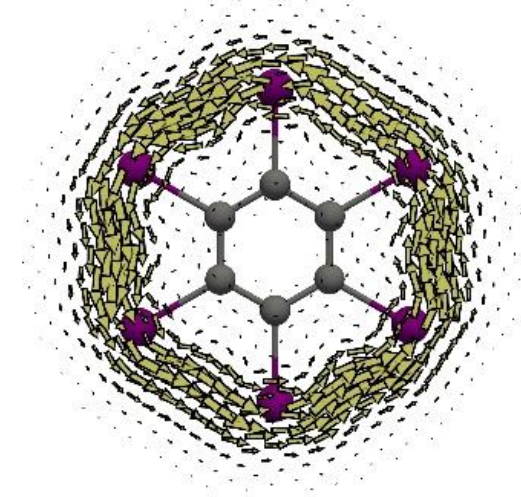
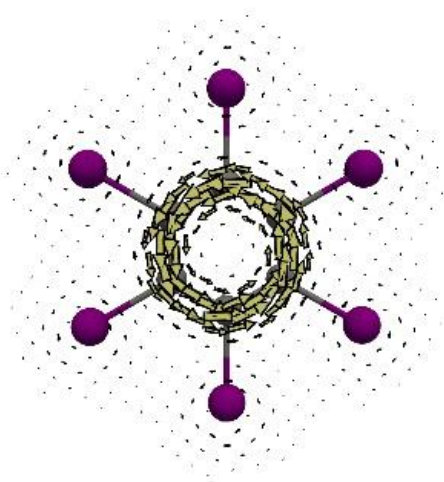
π -currents

σ -currents

C_6I_6



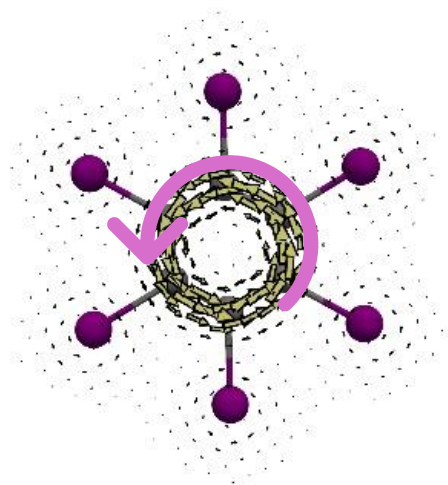
$C_6I_6^{2+}$



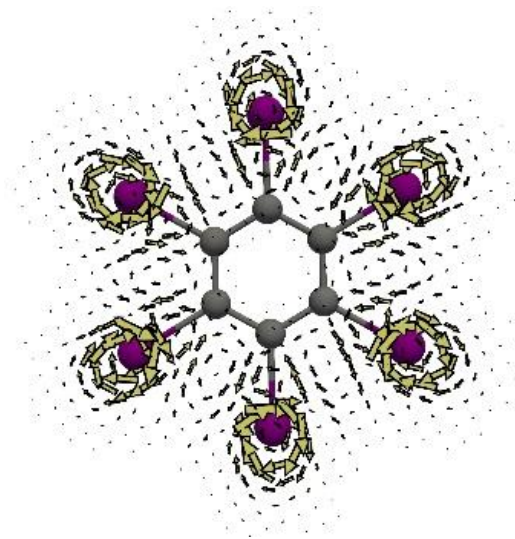
Periodo-benzene

π -currents

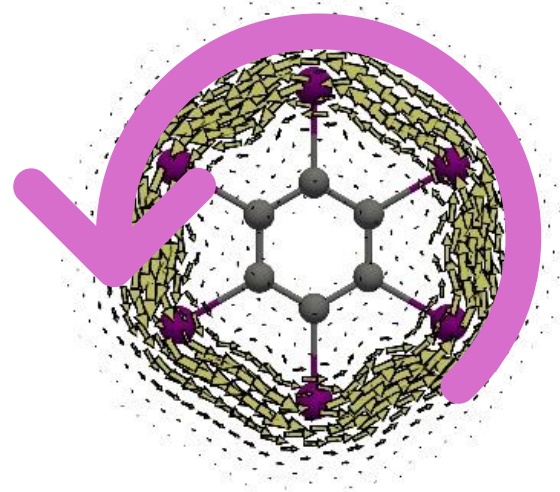
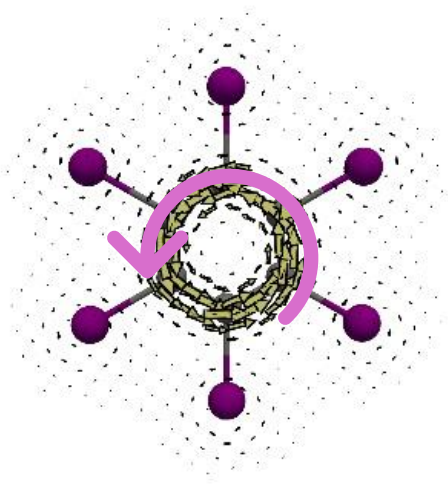
C_6I_6



σ -currents



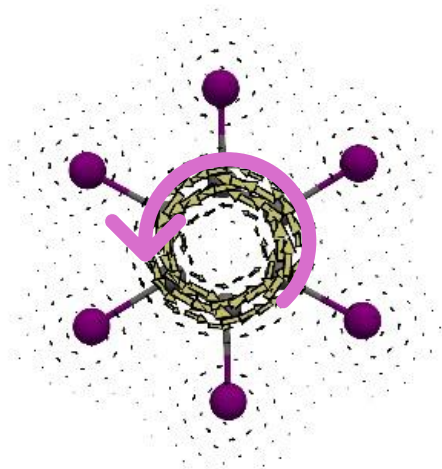
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Periodo-benzene

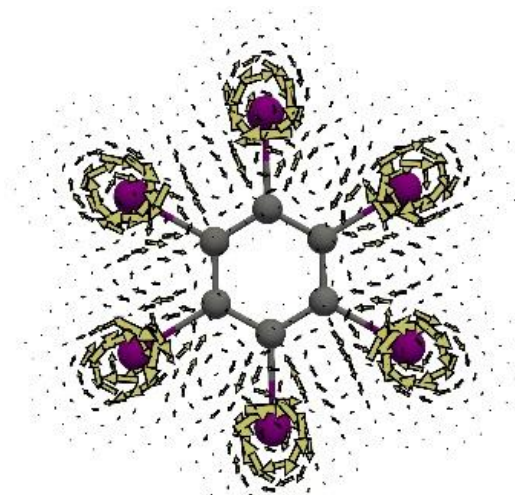
π -currents

C_6I_6



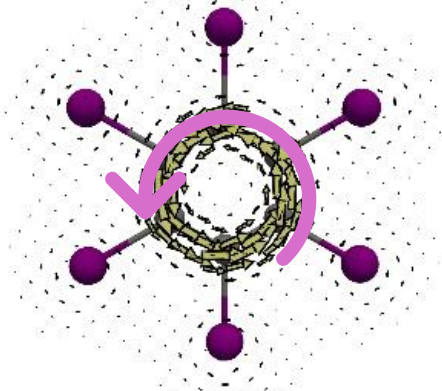
6 electrons

σ -currents

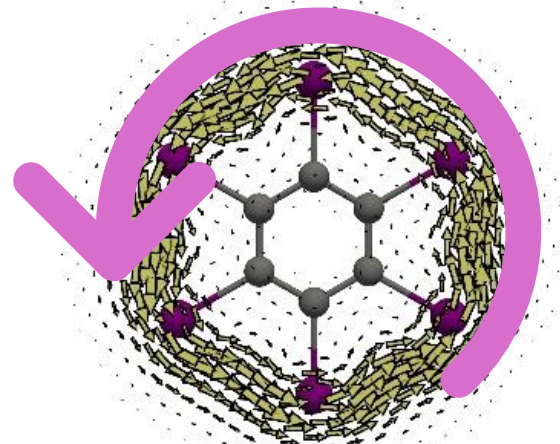


12 electrons

$C_6I_6^{2+}$

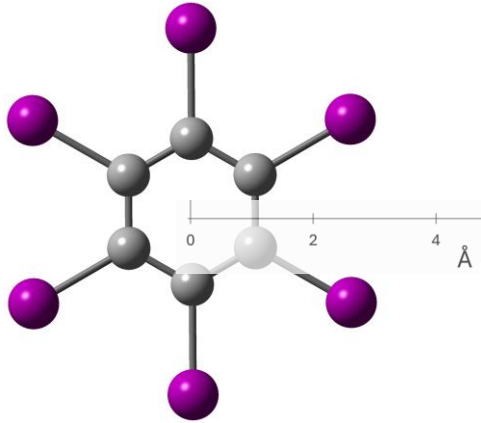


6 electrons

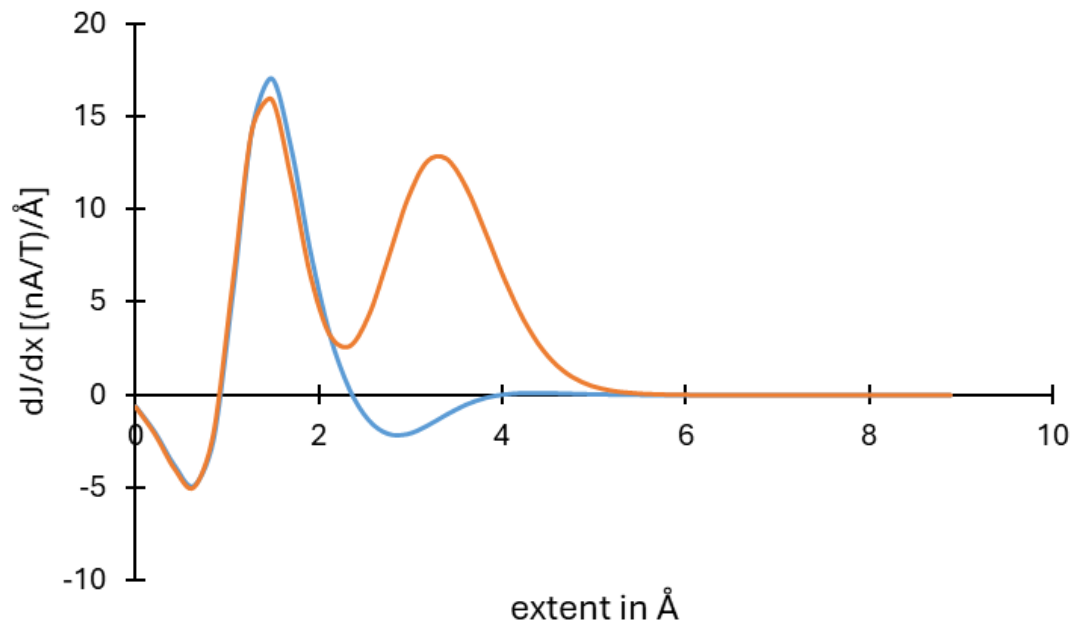
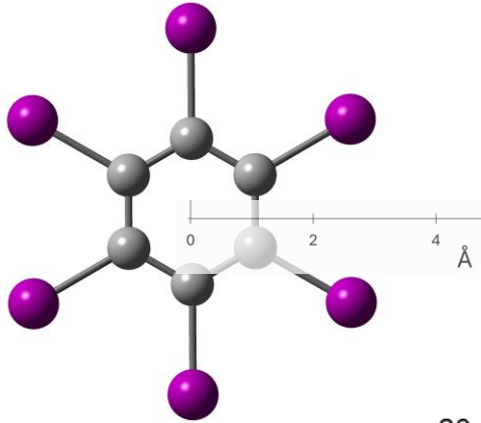


10 electrons

Periodo-benzene – bond current strengths profile



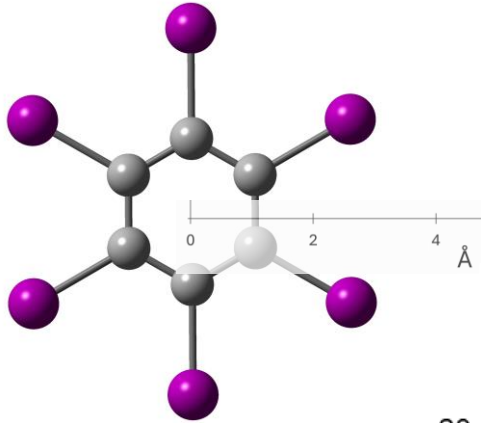
Periodo-benzene – bond current strengths profile



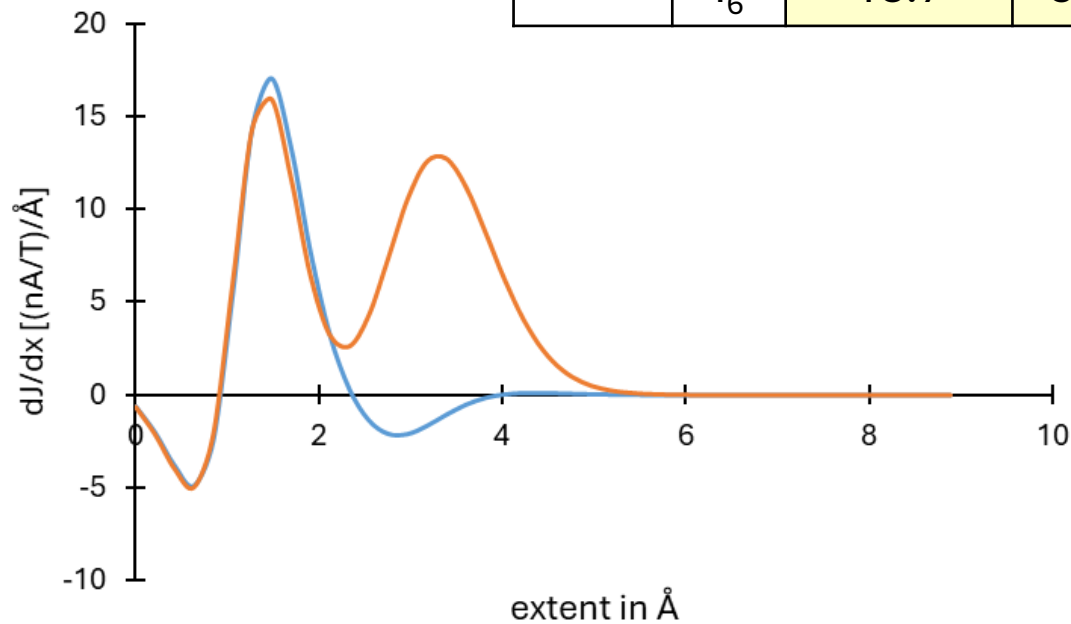
C_6I_6 – blue line

$C_6I_6^{2+}$ - orange line

Periodo-benzene – bond current strengths profile



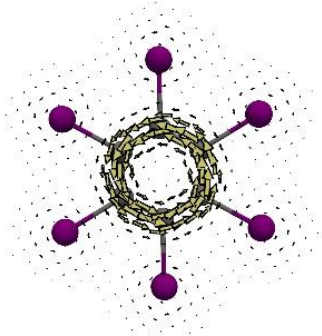
		J [nA T ⁻¹]	EDDB	I_{ring}
C_6I_6	C_6	9.4	5.074	0.625
	I_6	-1.0	0.227	0.065
$\text{C}_6\text{I}_6^{2+}$	C_6	8.6	5.038	0.622
	I_6	18.7	5.333	0.319



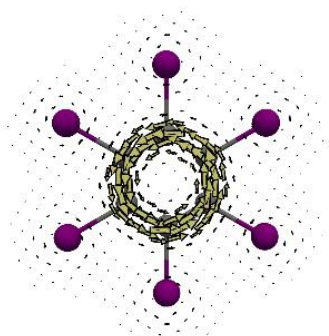
C₆I₆ – blue line

C₆I₆²⁺ – orange line

Periodo-benzene – π bond current strengths profile

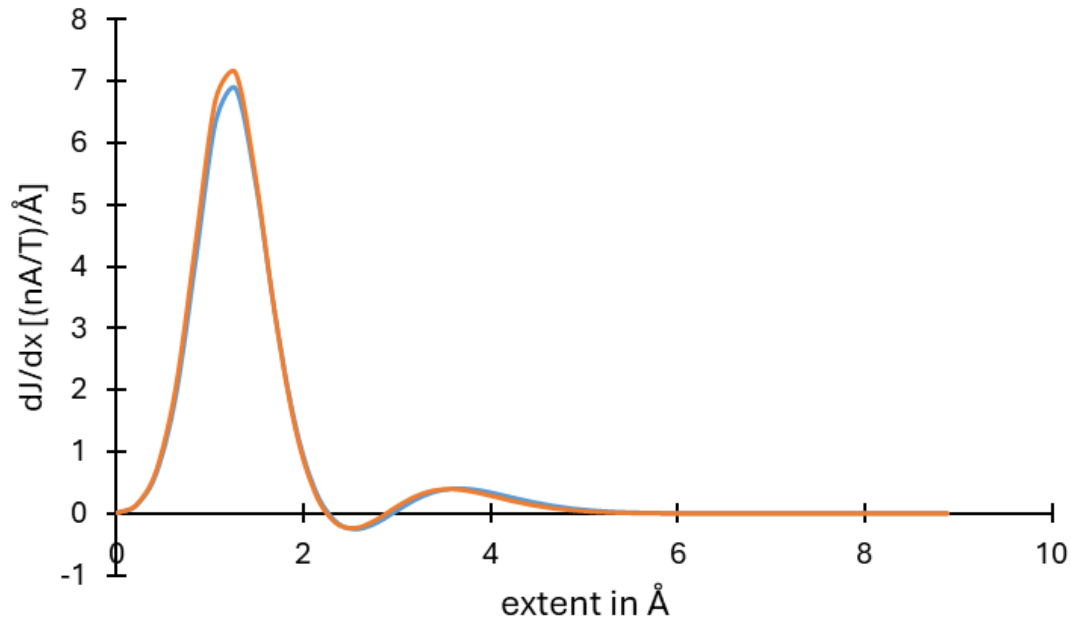
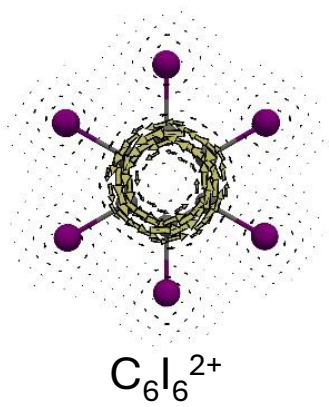
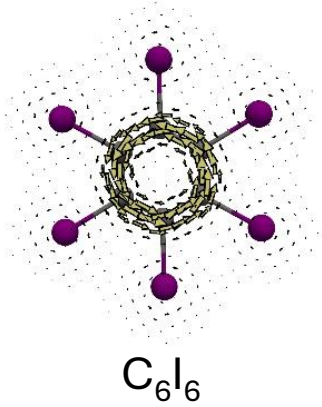


C_6I_6



$C_6I_6^{2+}$

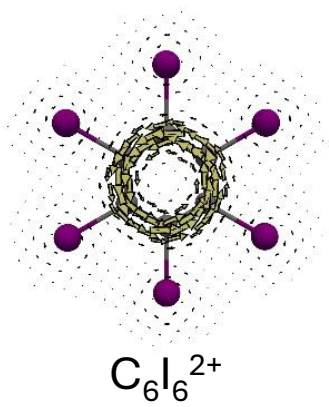
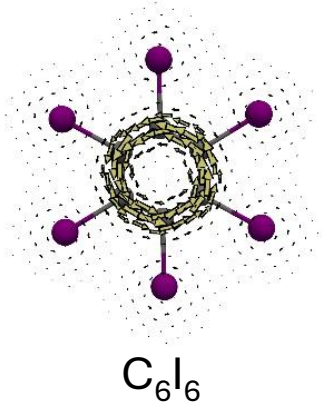
Periodo-benzene – π bond current strengths profile



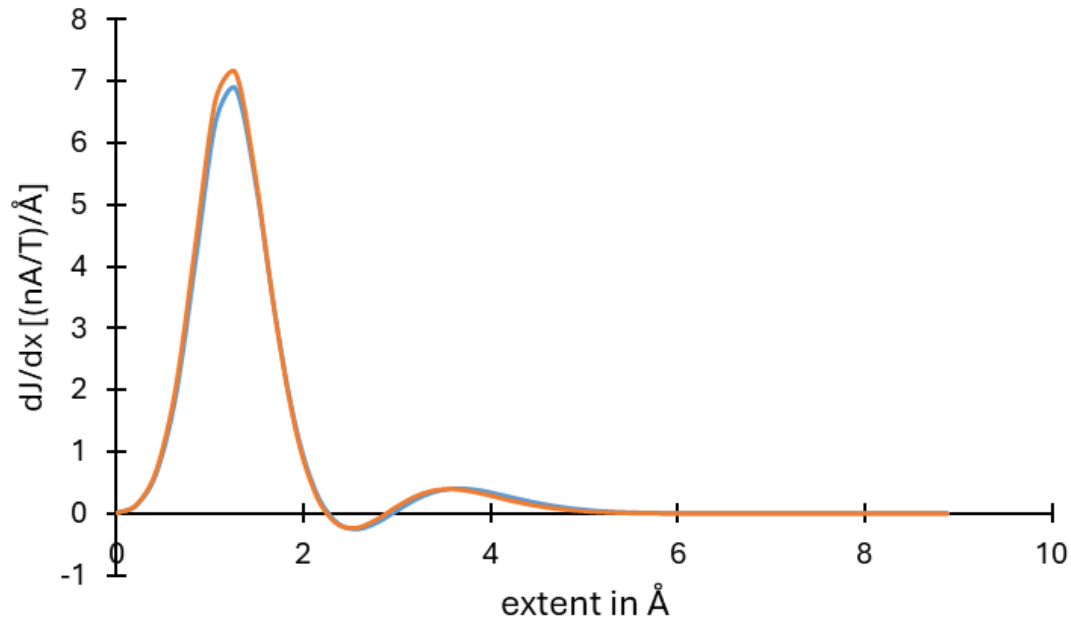
C_6I_6 – blue line

$C_6I_6^{2+}$ – orange line

Periodo-benzene – π bond current strengths profile



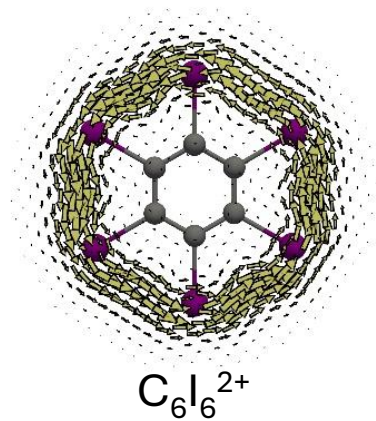
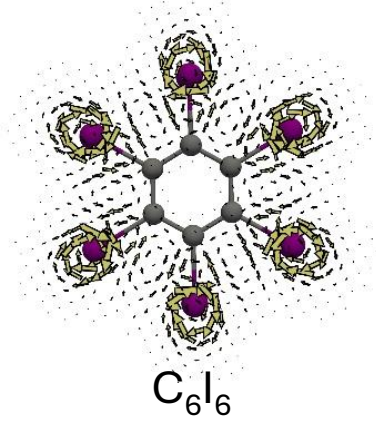
	J^π [nA T ⁻¹]	EDDB _{π}
C_6I_6	6.4	5.737
$C_6I_6^{2+}$	6.6	5.733



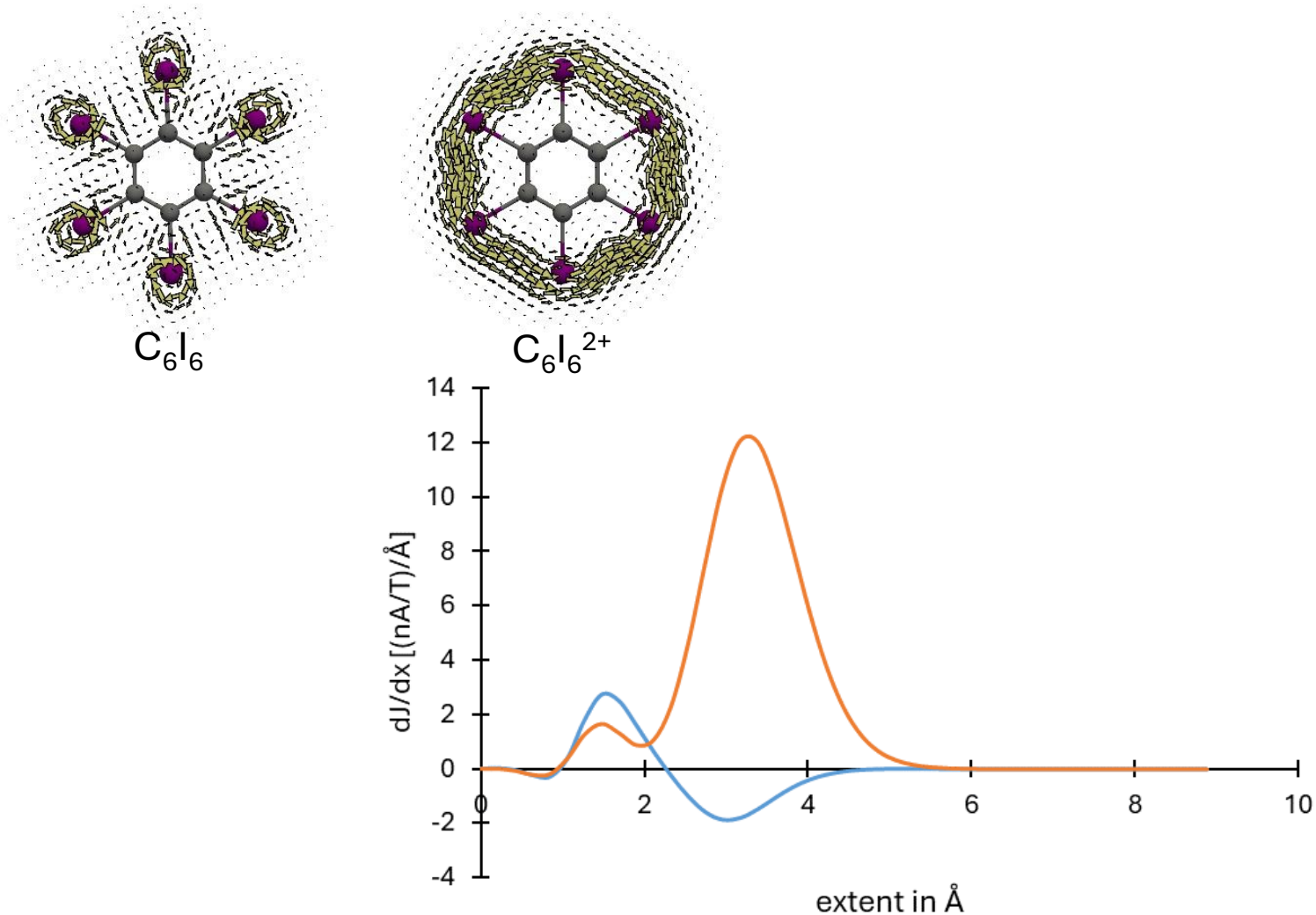
C_6I_6 – blue line

$C_6I_6^{2+}$ – orange line

Periodo-benzene – σ bond current strengths profile



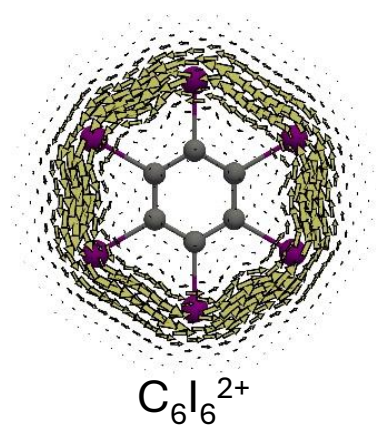
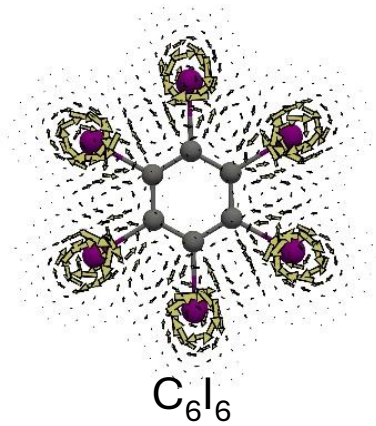
Periodo-benzene – σ bond current strengths profile



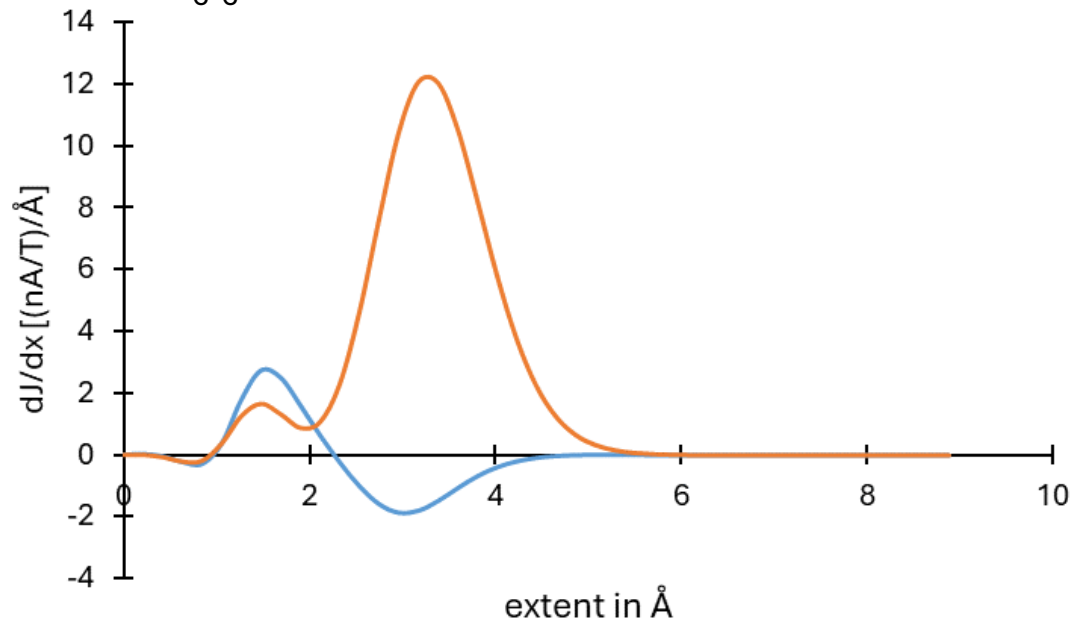
C_6I_6 – blue line

$C_6I_6^{2+}$ - orange line

Periodo-benzene – σ bond current strengths profile



	J^σ [nA T ⁻¹]	EDDB _{σ}
C_6I_6	-2.0	0.623
$C_6I_6^{2+}$	17.6	5.812

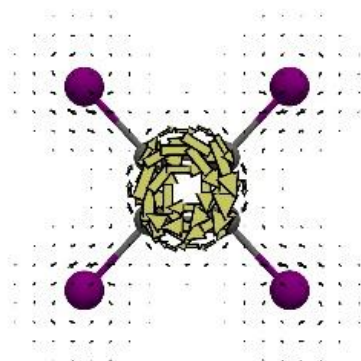


C_6I_6 – blue line

$C_6I_6^{2+}$ - orange line

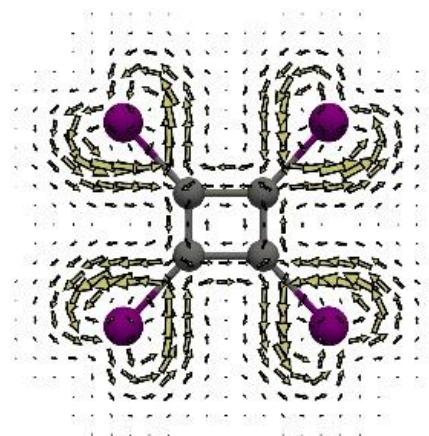
Periodo-cyclobutadiene

π -currents



C_4I_4

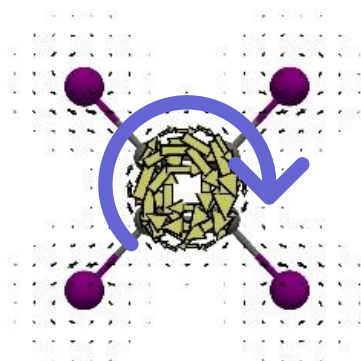
σ -currents



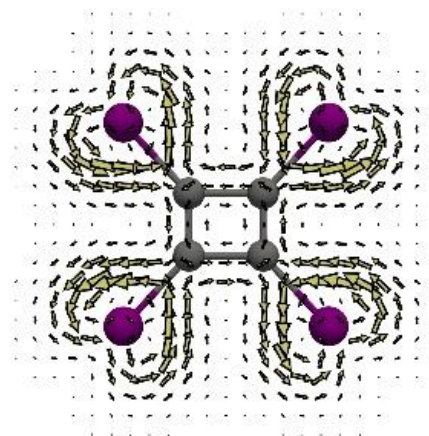
Periodo-cyclobutadiene

π -currents

C_4I_4



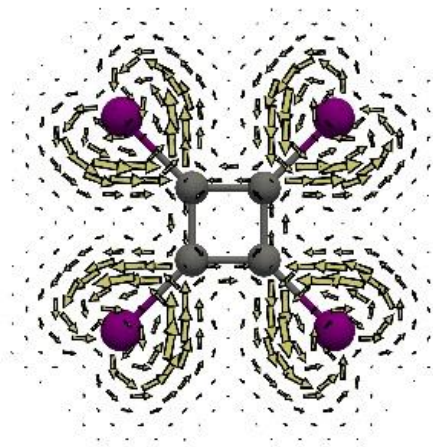
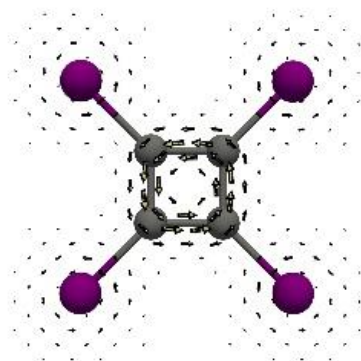
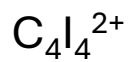
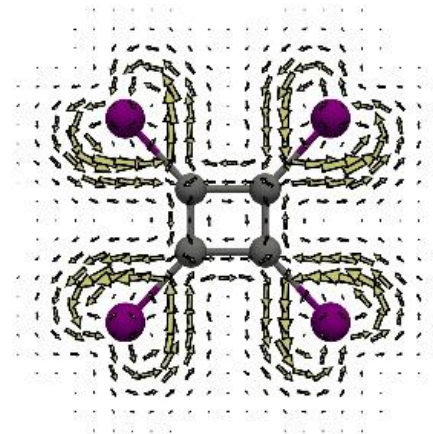
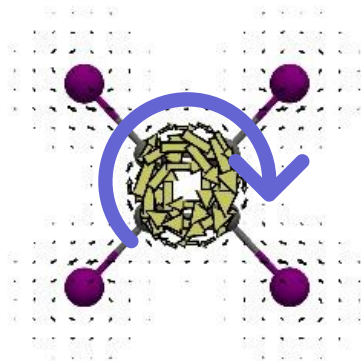
σ -currents



Periodo-cyclobutadiene

π -currents

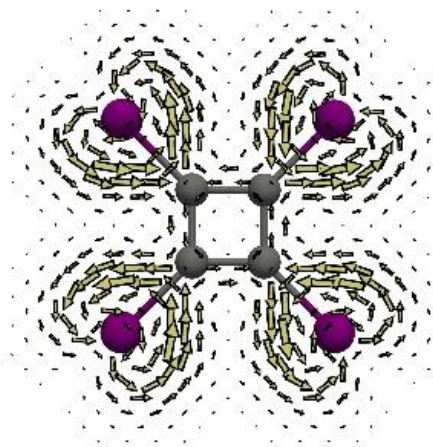
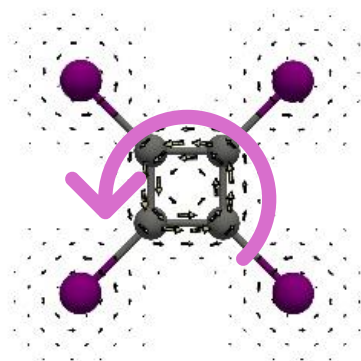
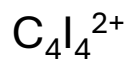
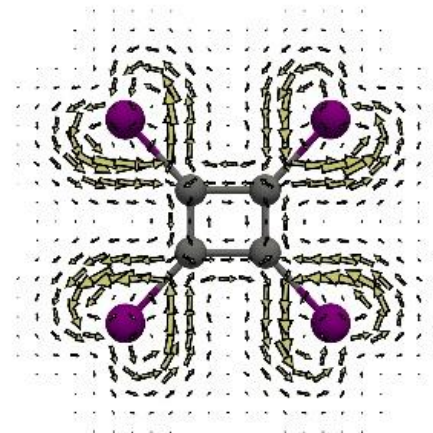
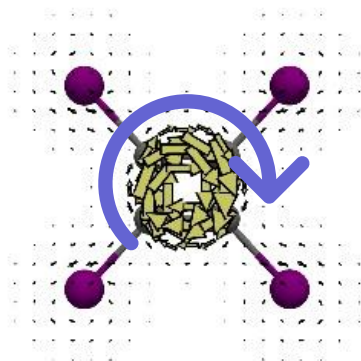
σ -currents



Periodo-cyclobutadiene

π -currents

σ -currents

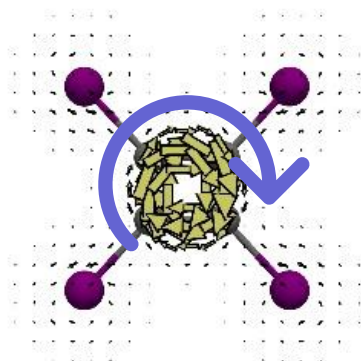


Periodo-cyclobutadiene

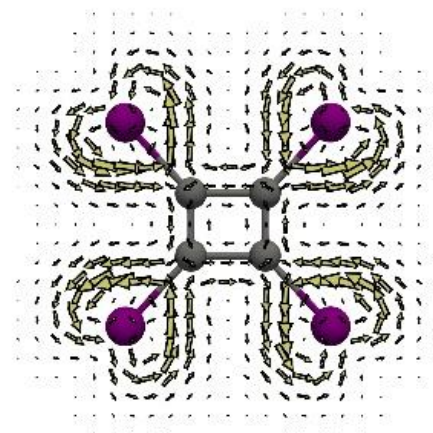
π -currents

σ -currents

C_4I_4

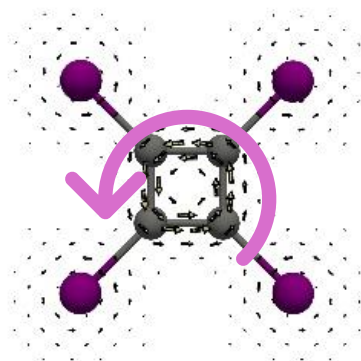


4 electrons

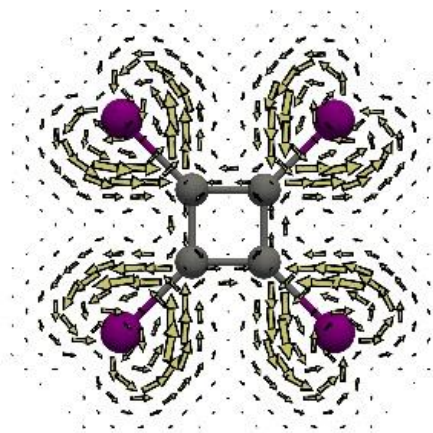


8 electrons

$C_4I_4^{2+}$

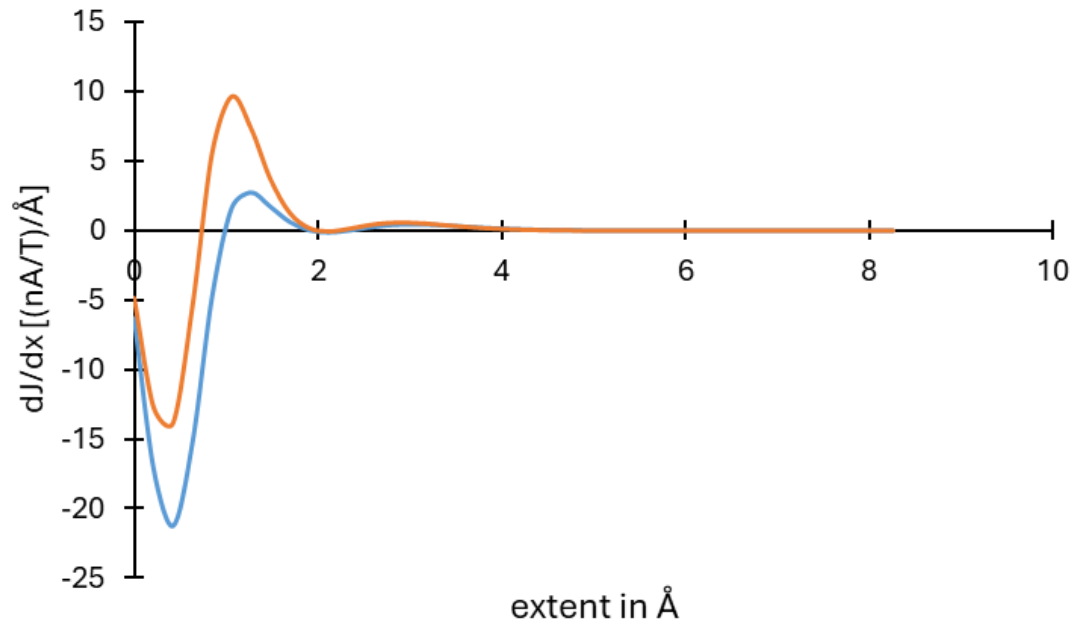
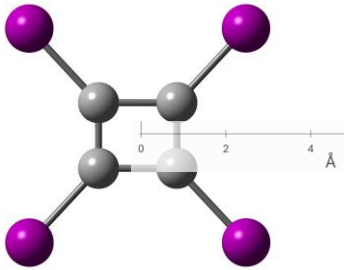


2 electrons



8 electrons

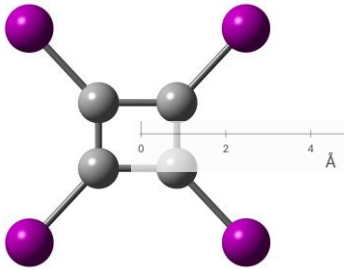
Periodo-cyclobutadiene – bond current strengths profile



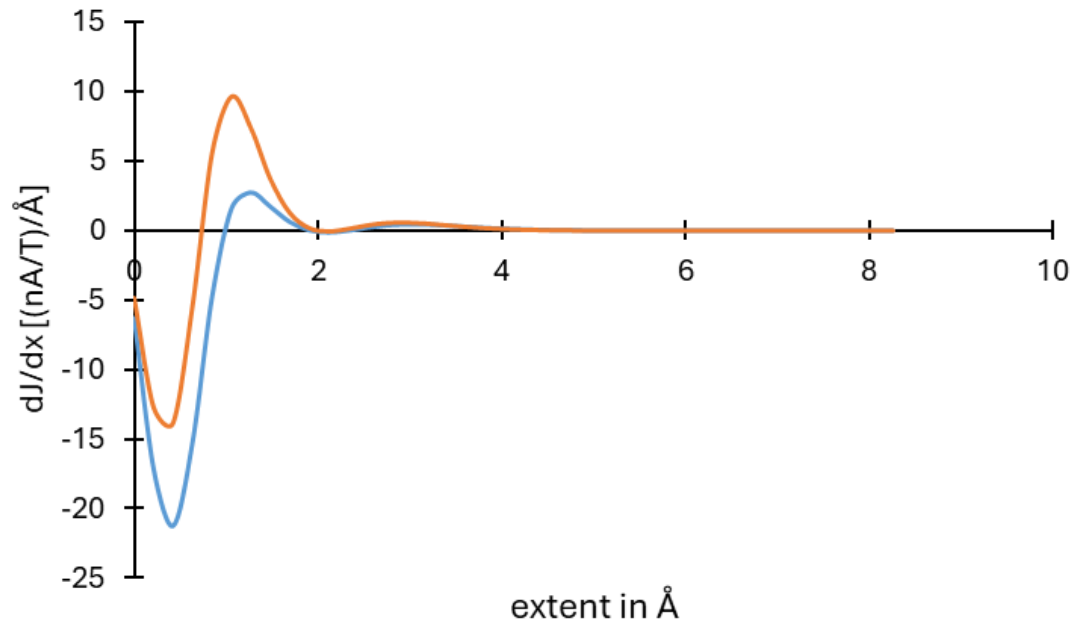
C₄I₄ – blue line

C₄I₄²⁺ – orange line

Periodo-cyclobutadiene – bond current strengths profile



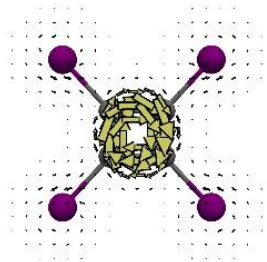
		$J \text{ [nA T}^{-1}\text{]}$	EDDB	I_{ring}
C_4I_4	C_4	-12.3	0.211	0.217
	I_4	0.5	0.073	0.072
$\text{C}_4\text{I}_4^{2+}$	C_4	-1.9	1.225	0.453
	I_4	0.6	0.064	0.050



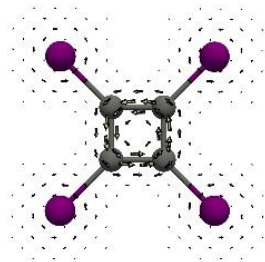
C_4I_4 – blue line

$\text{C}_4\text{I}_4^{2+}$ - orange line

Periodo-cyclobutadiene – π bond current strengths profile

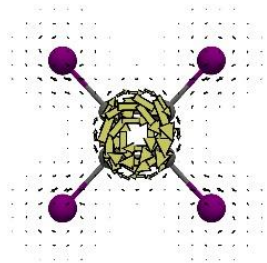


C_4I_4

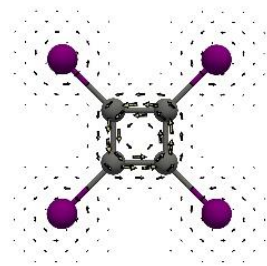


$C_4I_4^{2+}$

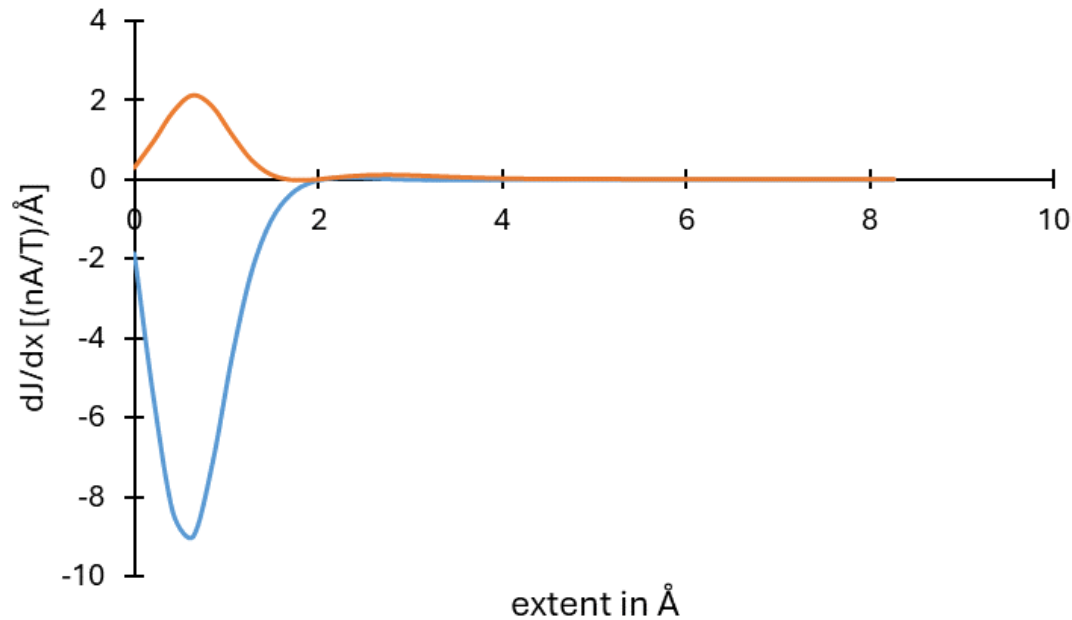
Periodo-cyclobutadiene – π bond current strengths profile



C_4I_4



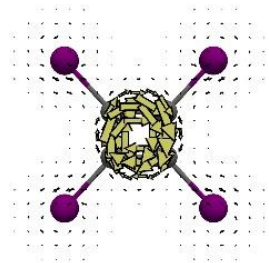
$C_4I_4^{2+}$



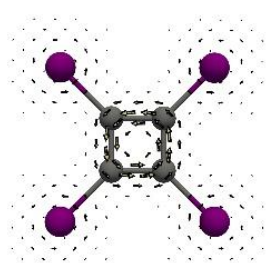
C_4I_4 – blue line

$C_4I_4^{2+}$ – orange line

Periodo-cyclobutadiene – π bond current strengths profile

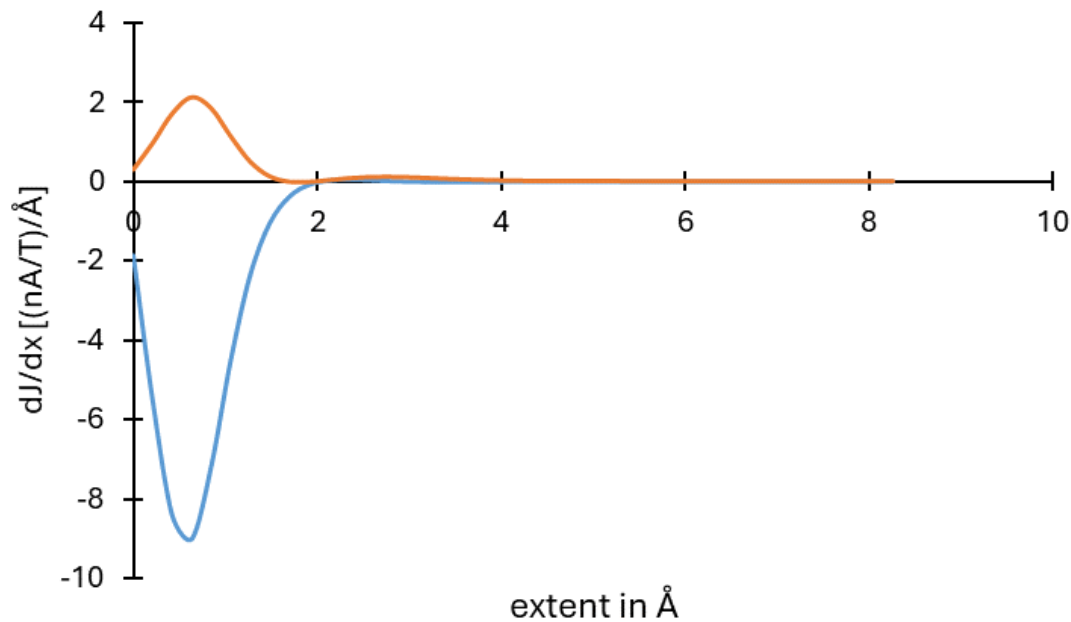


C_4I_4



$C_4I_4^{2+}$

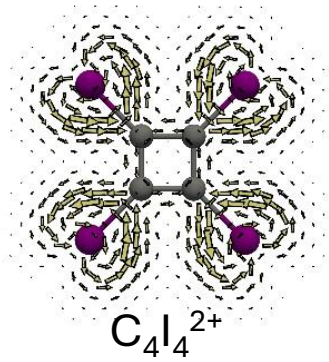
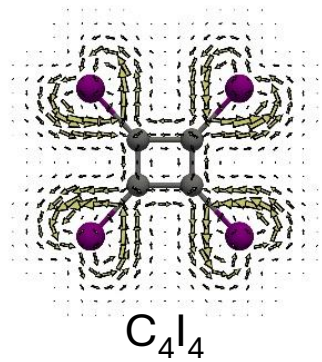
	J^π [nA T ⁻¹]	EDDB _{π}
C_4I_4	-8.6	0.719
$C_4I_4^{2+}$	1.8	1.451



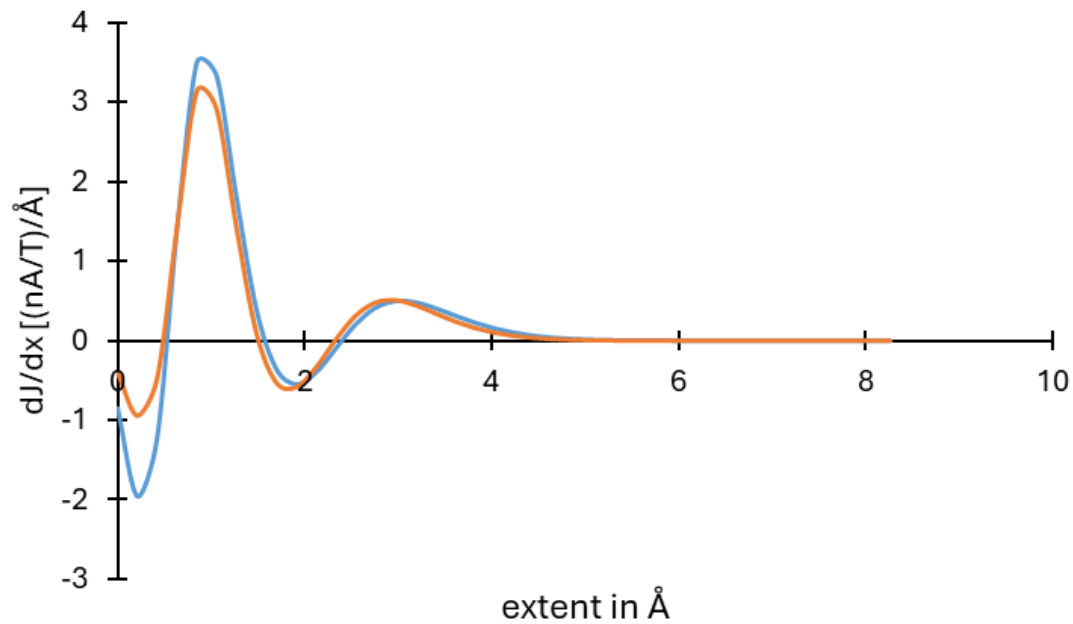
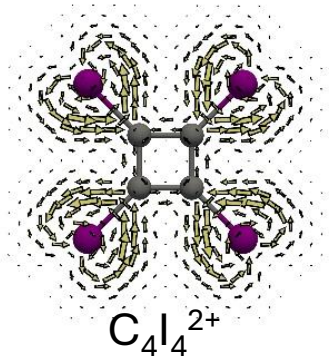
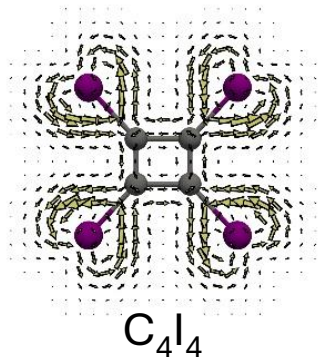
C_4I_4 – blue line

$C_4I_4^{2+}$ - orange line

Periodo-cyclobutadiene – σ bond current strengths profile



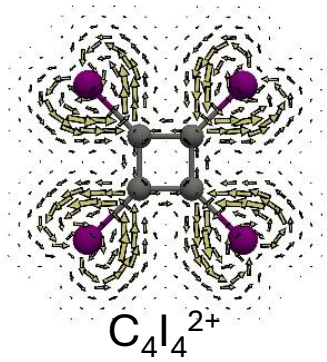
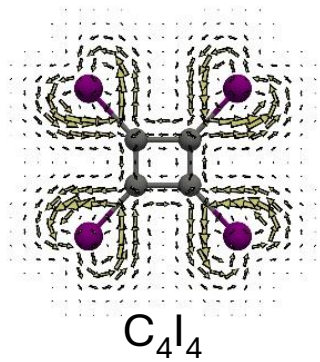
Periodo-cyclobutadiene – σ bond current strengths profile



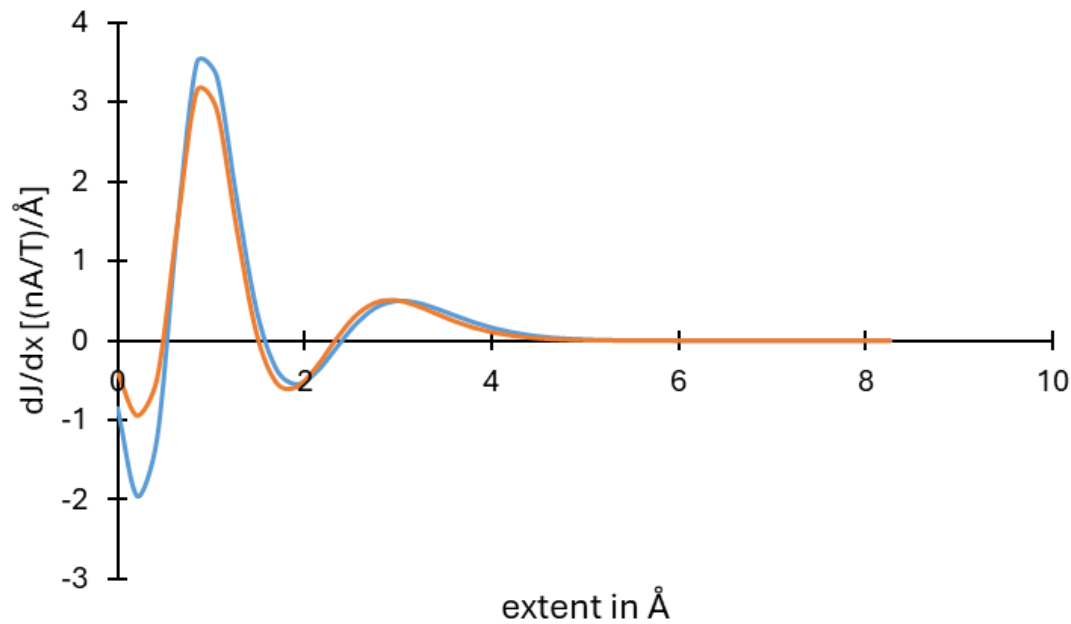
C_4I_4 – blue line

$C_4I_4^{2+}$ - orange line

Periodo-cyclobutadiene – σ bond current strengths profile



	J^σ [nA T ⁻¹]	EDDB _{σ}
C_4I_4	0.5	0.558
$C_4I_4^{2+}$	0.5	0.698



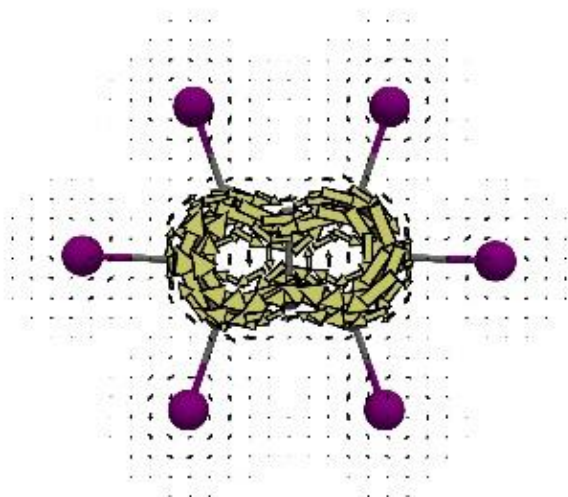
C_4I_4 – blue line

$C_4I_4^{2+}$ - orange line

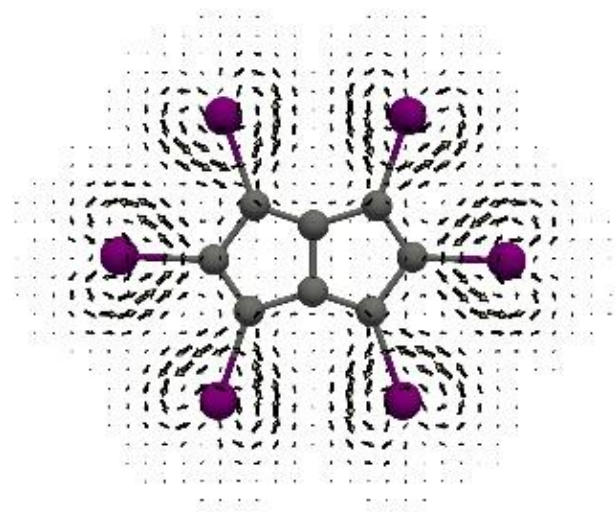
Periodo-pentalene

π -currents

C_8I_6



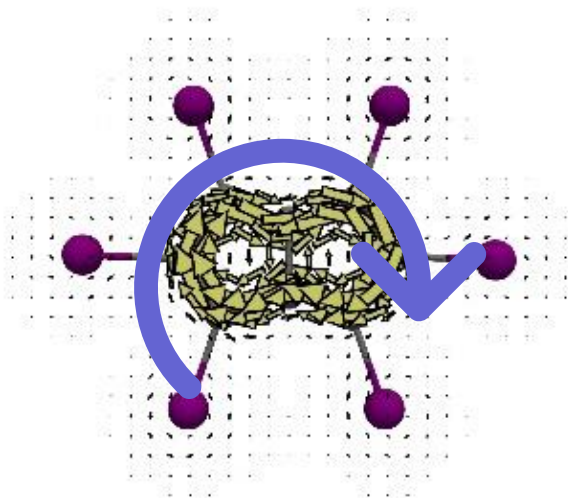
σ -currents



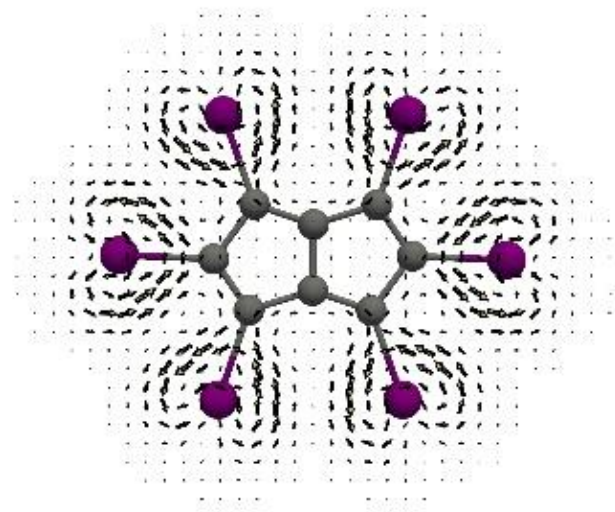
Periodo-pentalene

π -currents

C_8I_6



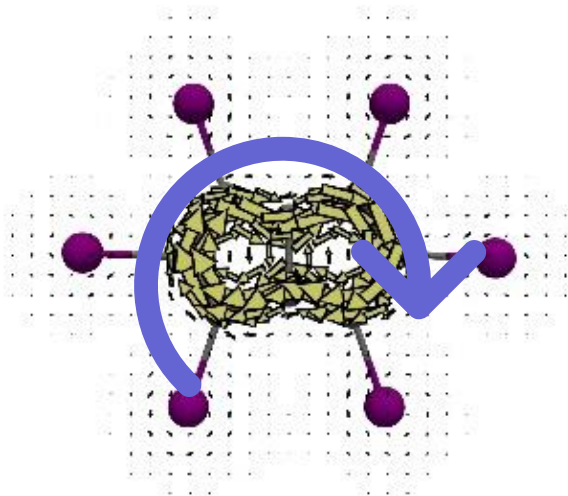
σ -currents



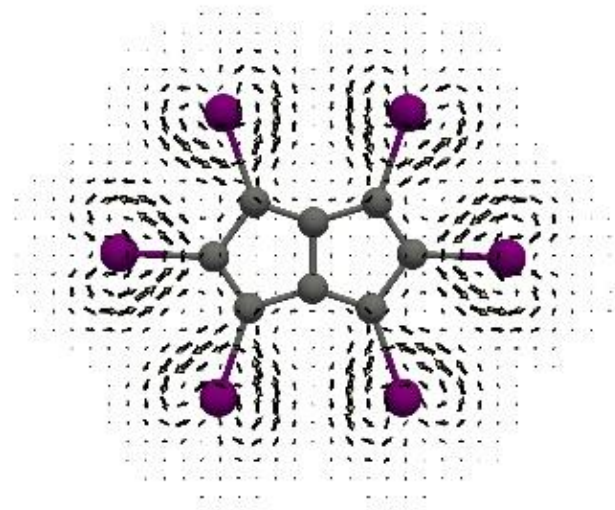
Periodo-pentalene

π -currents

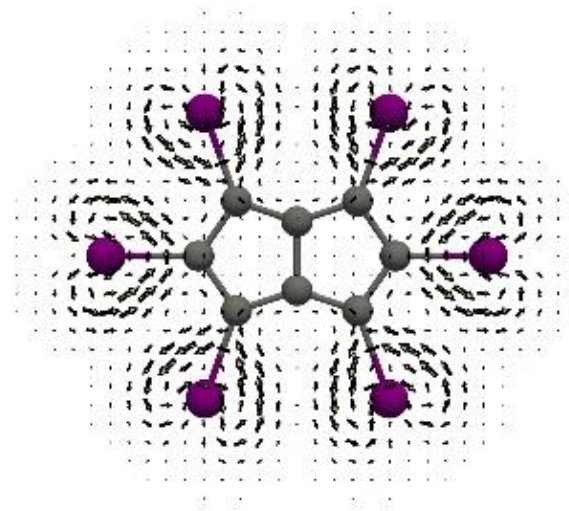
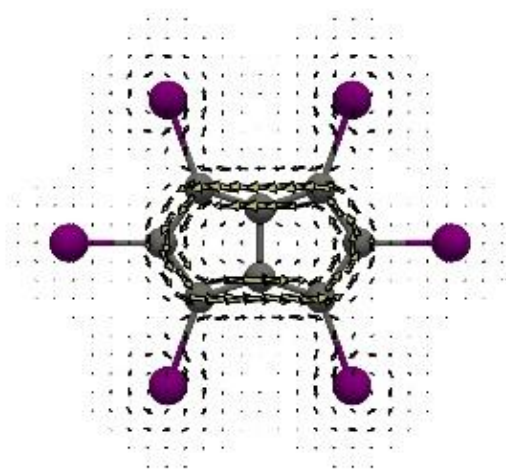
C_8I_6



σ -currents



$C_8I_6^{2+}$

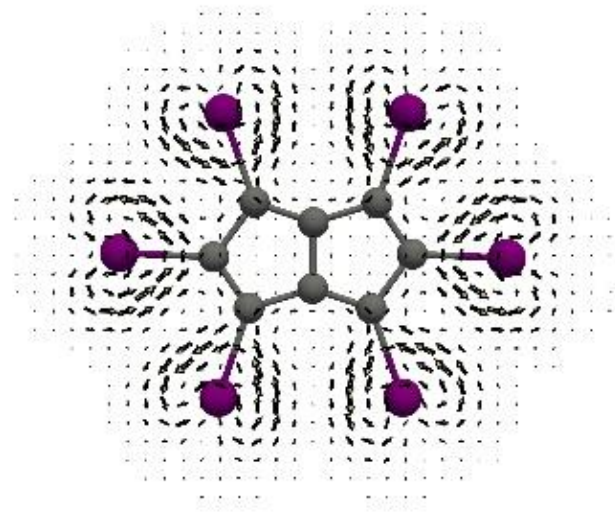
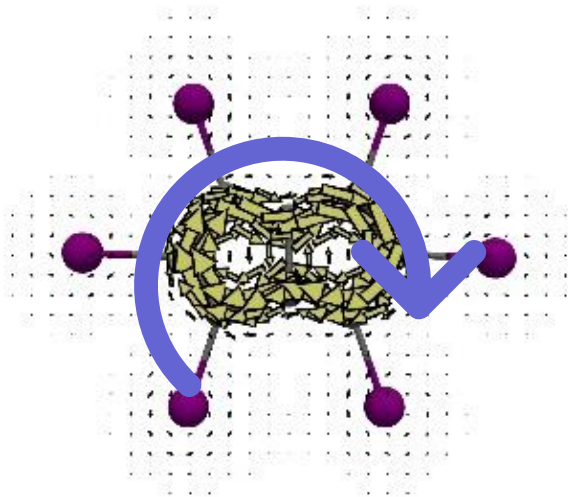


Periodo-pentalene

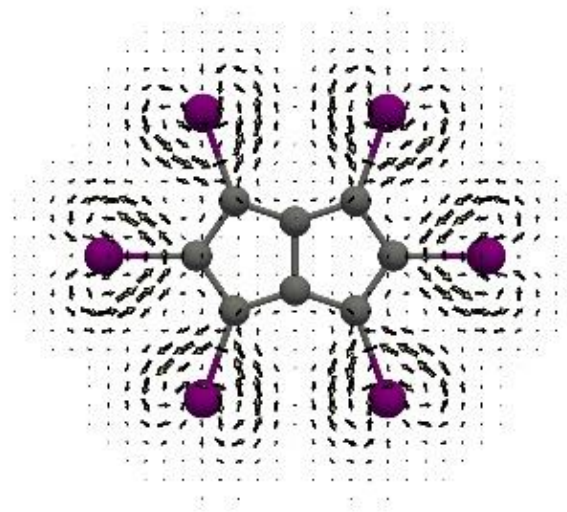
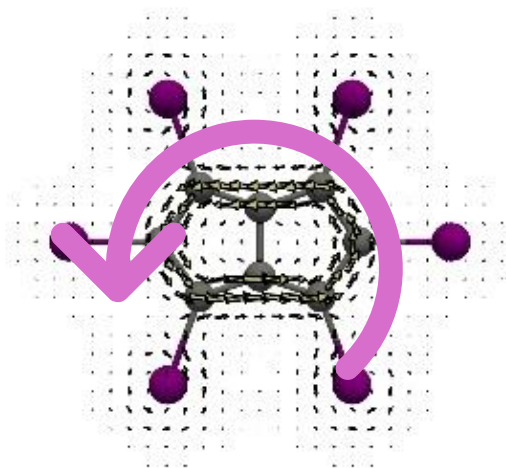
π -currents

σ -currents

C_8I_6



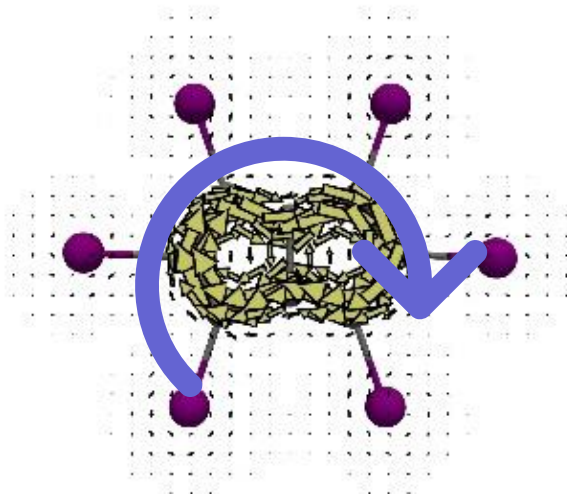
$C_8I_6^{2+}$



Periodo-pentalene

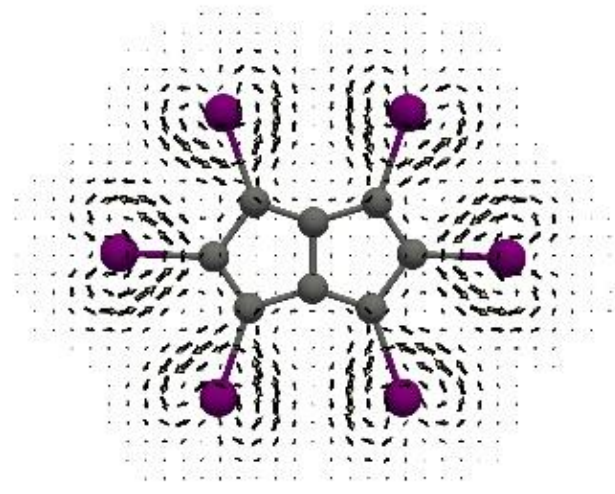
π -currents

C_8I_6



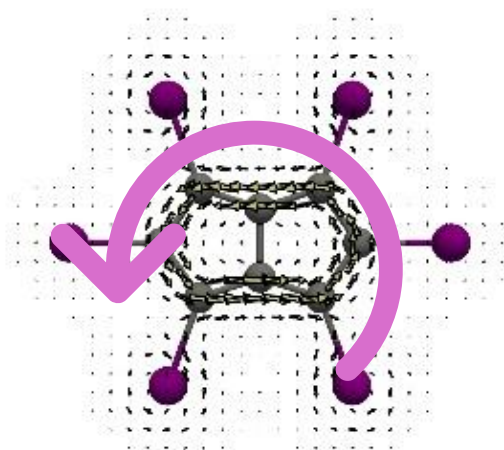
8 electrons

σ -currents

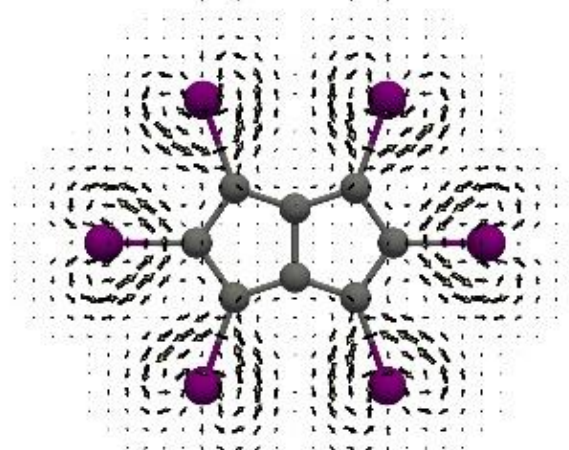


12 electrons

$C_8I_6^{2+}$

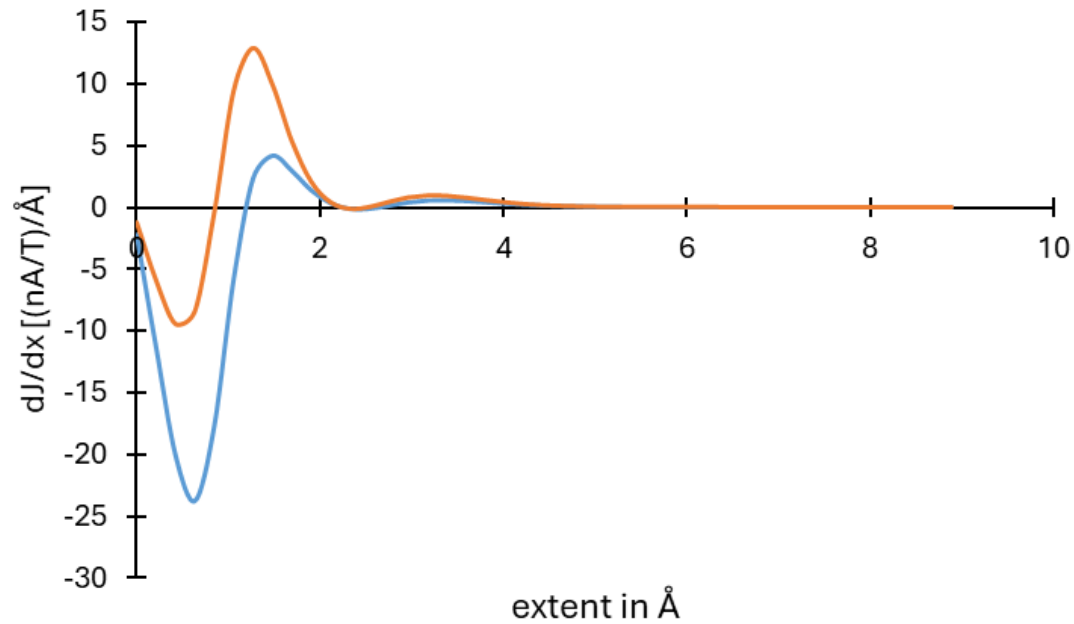
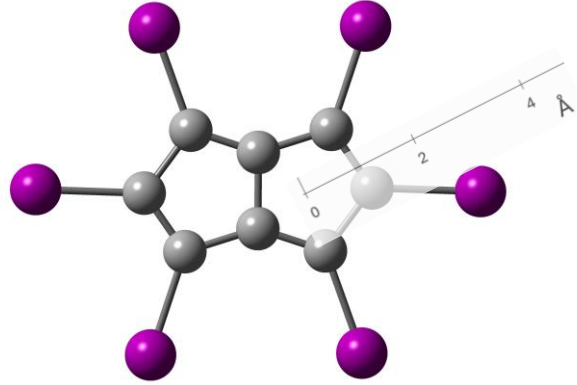


6 electrons



12 electrons

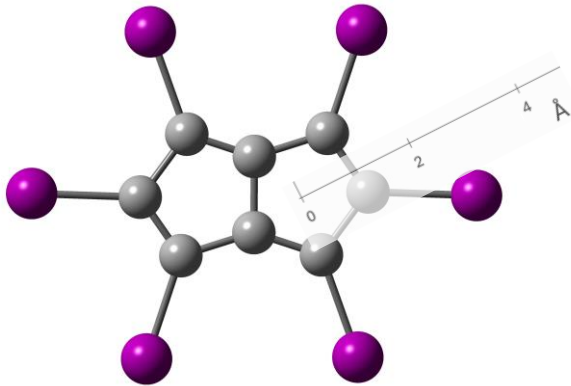
Periodo-pentalene – bond current strengths profile



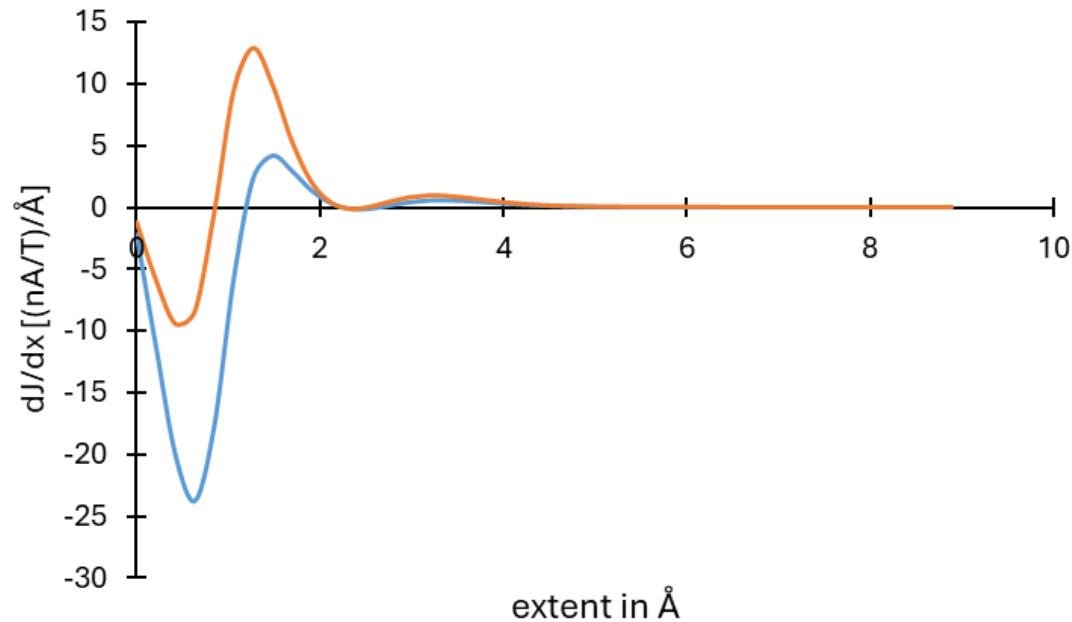
C_8I_6 – blue line

$C_8I_6^{2+}$ - orange line

Periodo-pentalene – bond current strengths profile



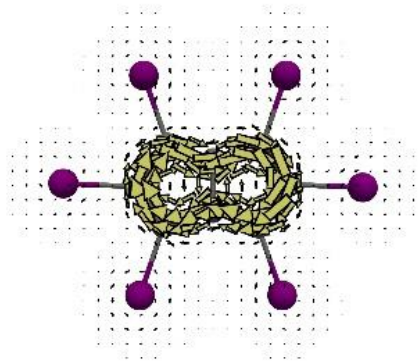
		$J [\text{nA T}^{-1}]$	EDDB	I_{ring}
C_8I_6	C_8	-14.8	2.511	0.516
	I_6	0.5	0.214	0.051
$\text{C}_8\text{I}_6^{2+}$	C_8	3.1	4.481	0.548
	I_6	1.0	0.530	0.043



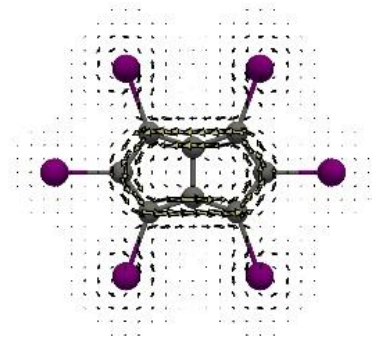
C_8I_6 – blue line

$\text{C}_8\text{I}_6^{2+}$ - orange line

Periodo-pentalene – π bond current strengths profile

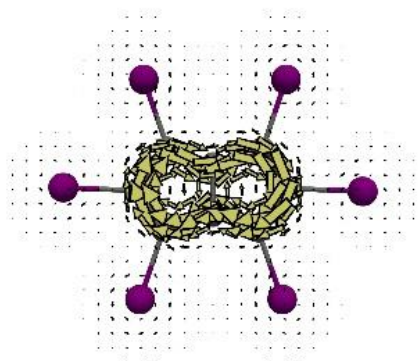


C_8I_6

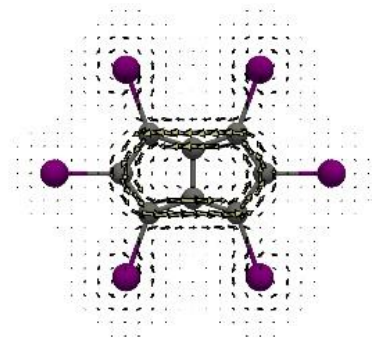


$C_8I_6^{2+}$

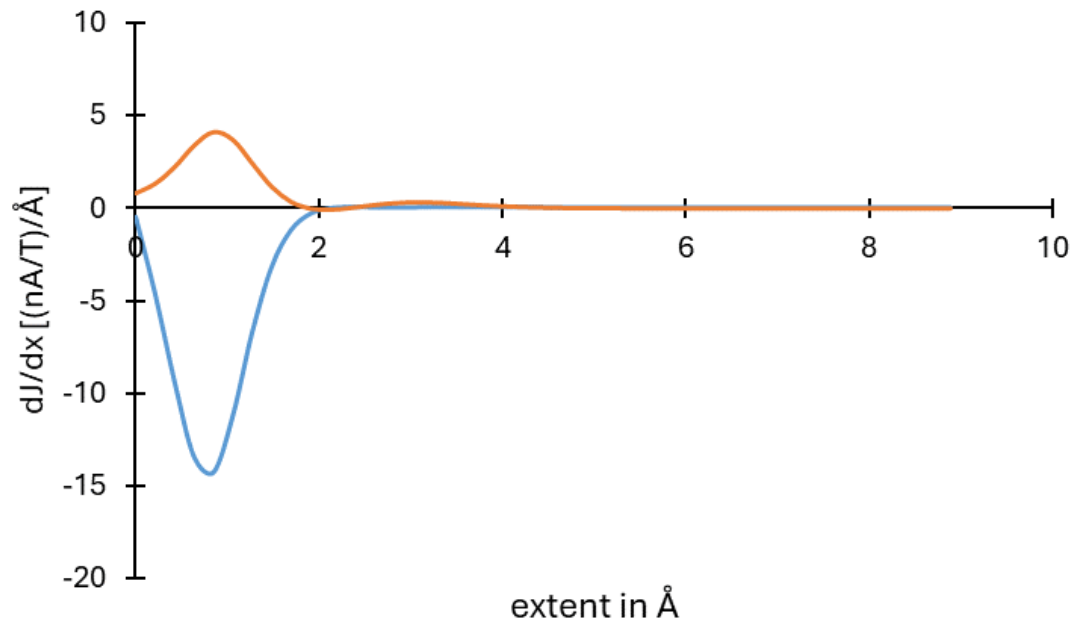
Periodo-pentalene – π bond current strengths profile



C_8I_6



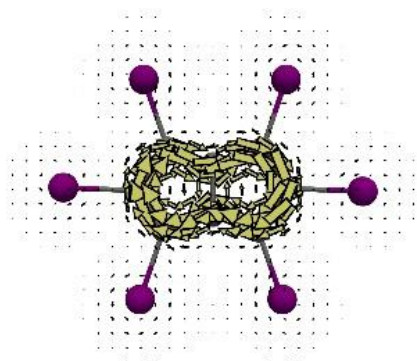
$C_8I_6^{2+}$



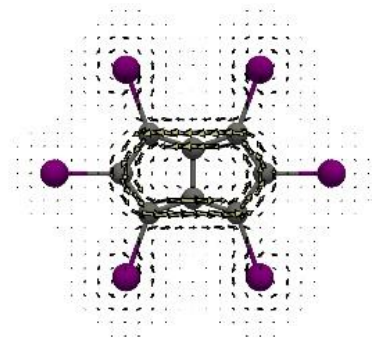
C_8I_6 – blue line

$C_8I_6^{2+}$ - orange line

Periodo-pentalene – π bond current strengths profile

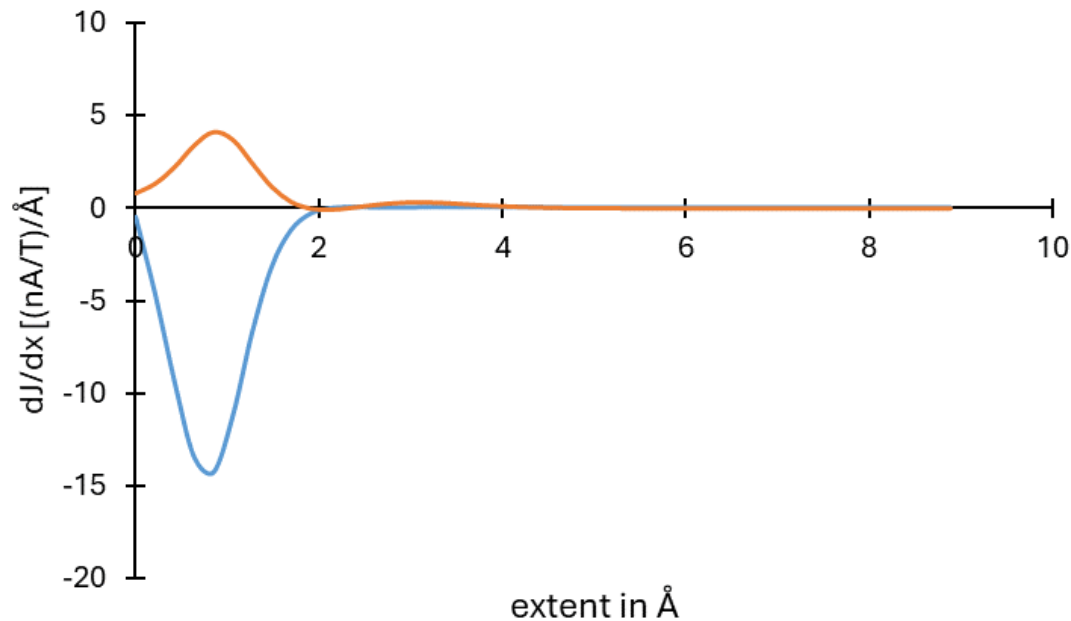


C_8I_6



$C_8I_6^{2+}$

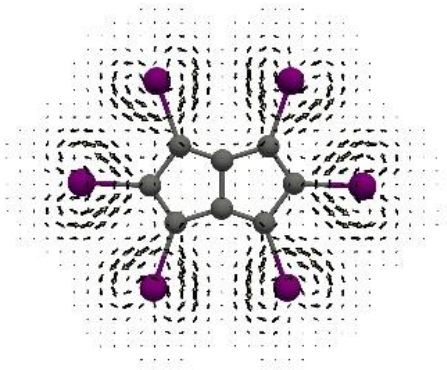
	J^π [nA T ⁻¹]	EDDB _{π}
C_8I_6	-13.8	3.297
$C_8I_6^{2+}$	4.1	5.340



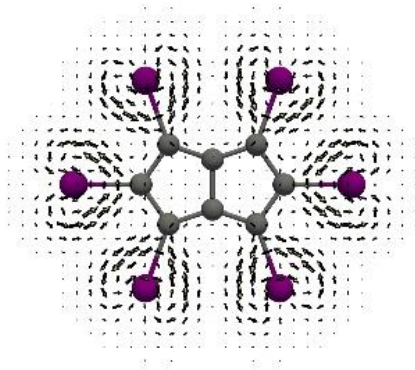
C_8I_6 – blue line

$C_8I_6^{2+}$ - orange line

Periodo-pentalene – σ bond current strengths profile

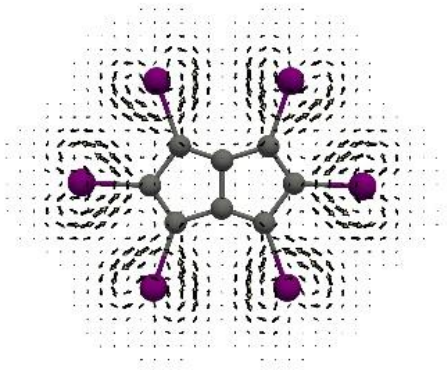


C_8I_6

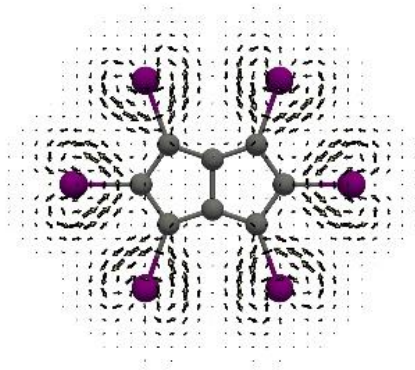


$C_8I_6^{2+}$

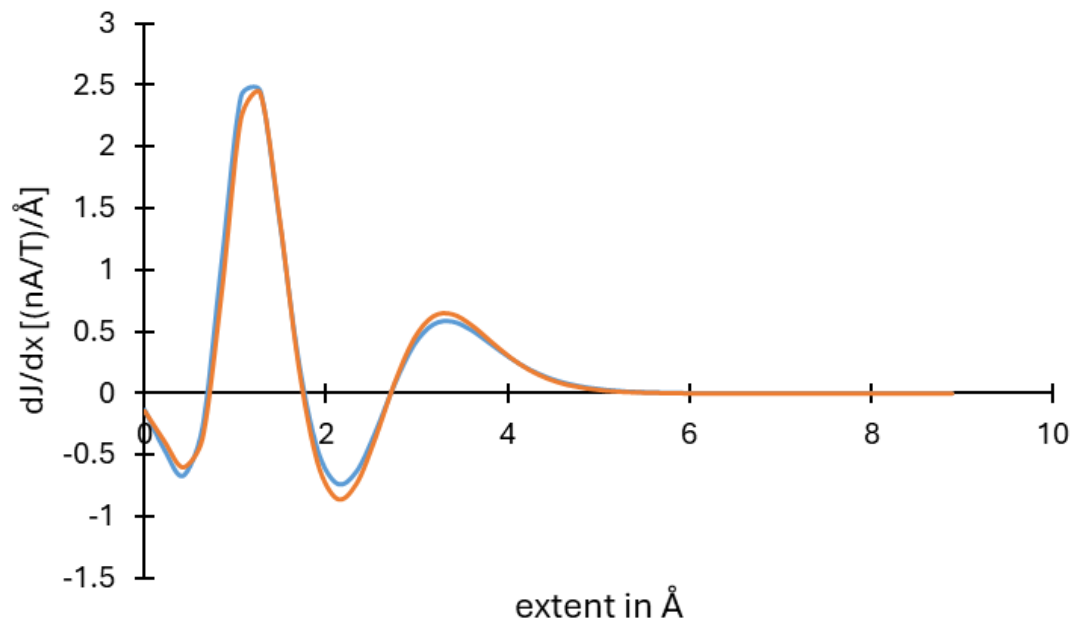
Periodo-pentalene – σ bond current strengths profile



C_8I_6



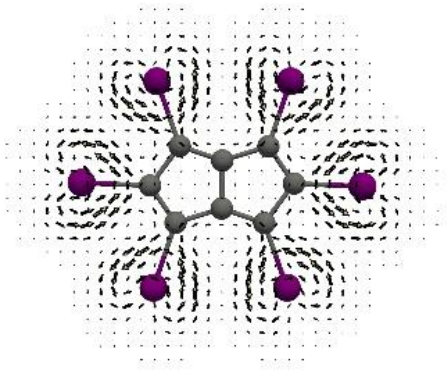
$C_8I_6^{2+}$



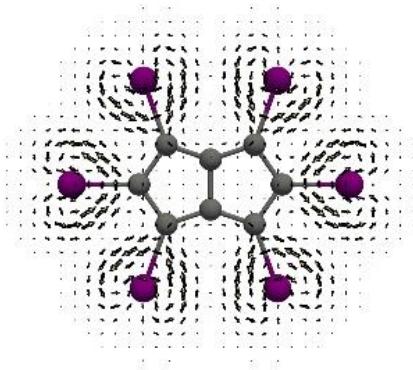
C_8I_6 – blue line

$C_8I_6^{2+}$ - orange line

Periodo-pentalene – σ bond current strengths profile

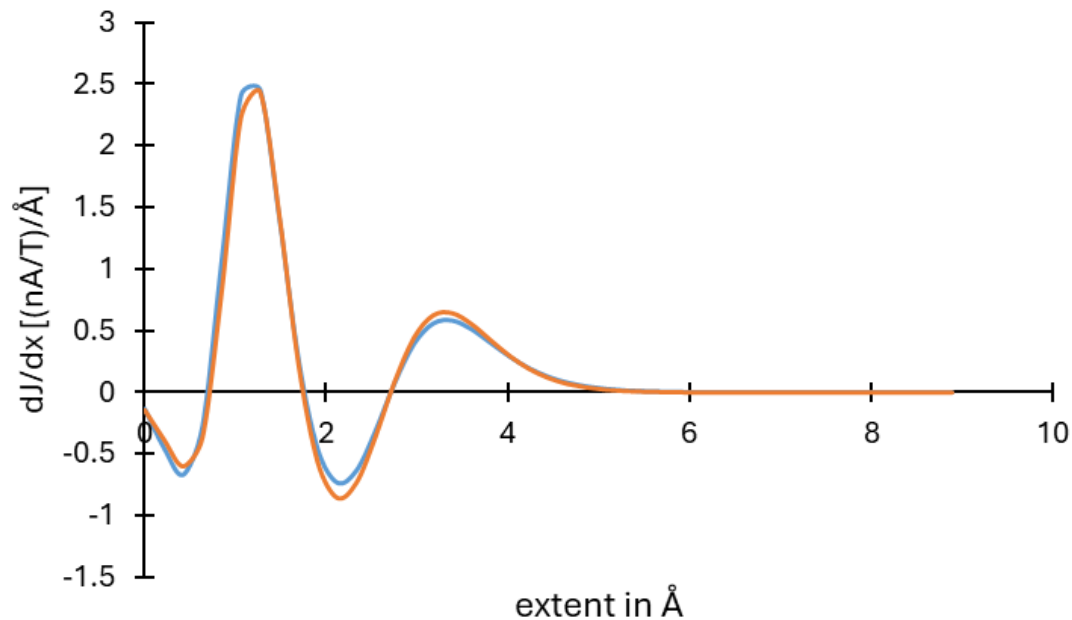


C_8I_6



$C_8I_6^{2+}$

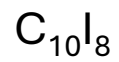
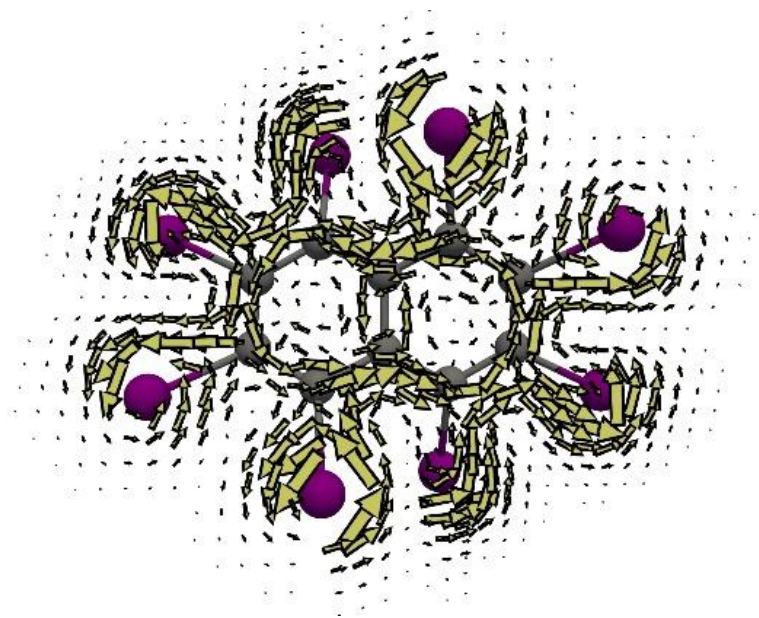
	$J^\sigma [\text{nA T}^{-1}]$	EDDB_σ
C_8I_6	1.0	0.609
$C_8I_6^{2+}$	0.9	0.885



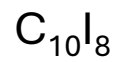
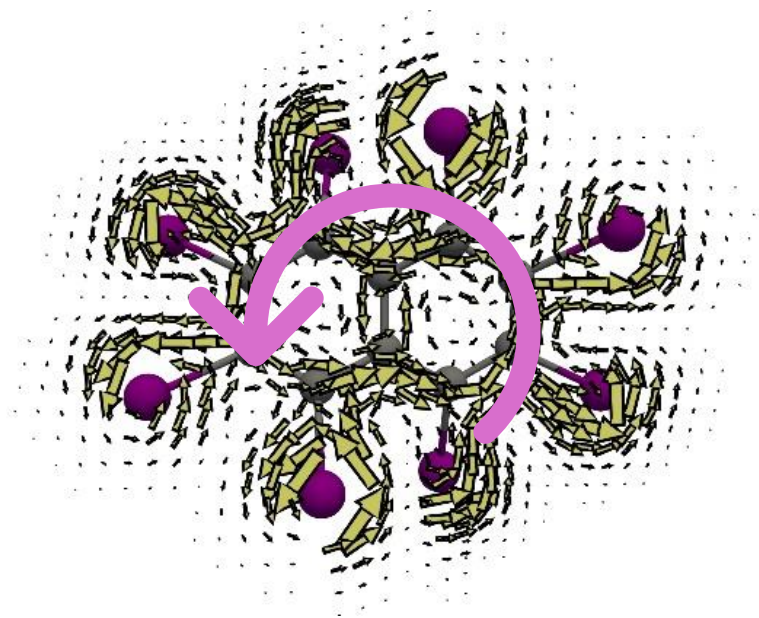
C_8I_6 – blue line

$C_8I_6^{2+}$ – orange line

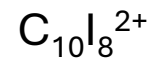
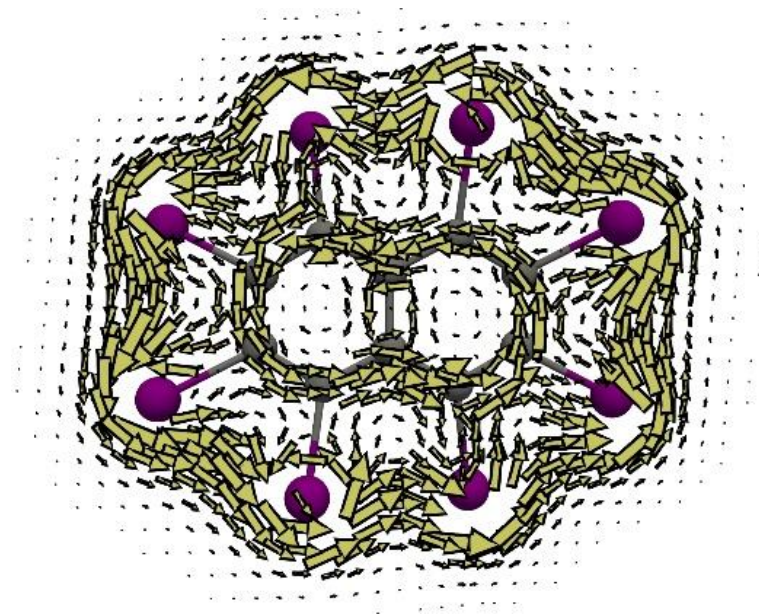
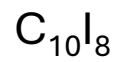
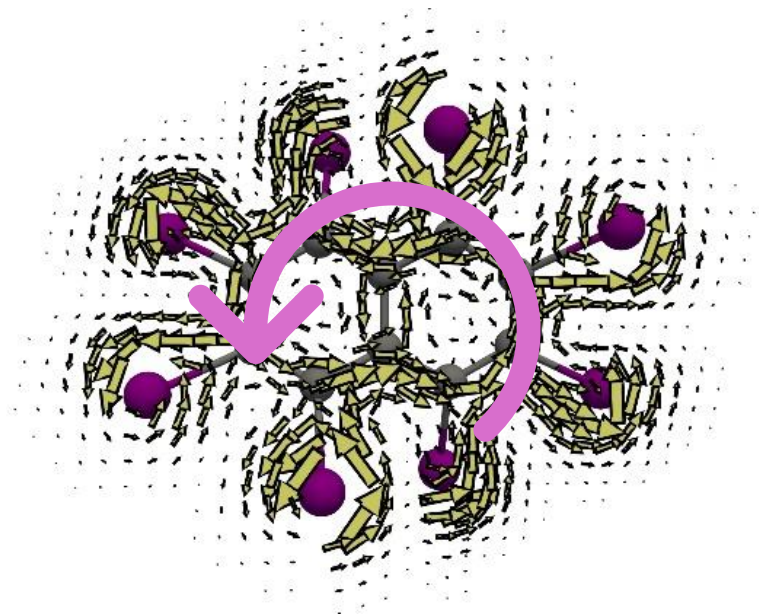
Periodo-naphtalene



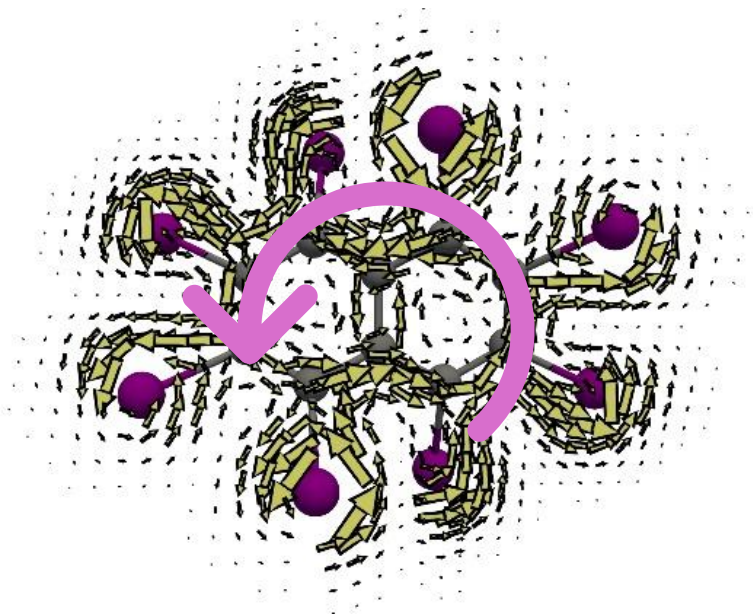
Periodo-naphtalene



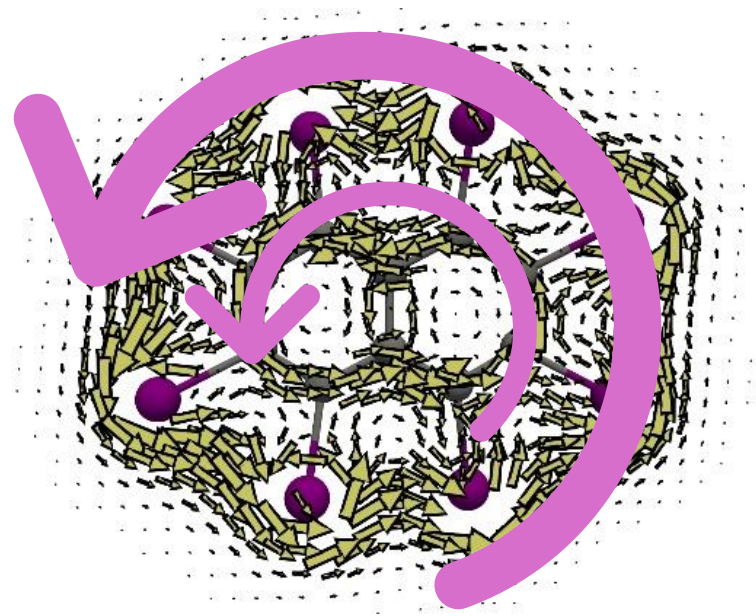
Periodo-naphtalene



Periodo-naphtalene

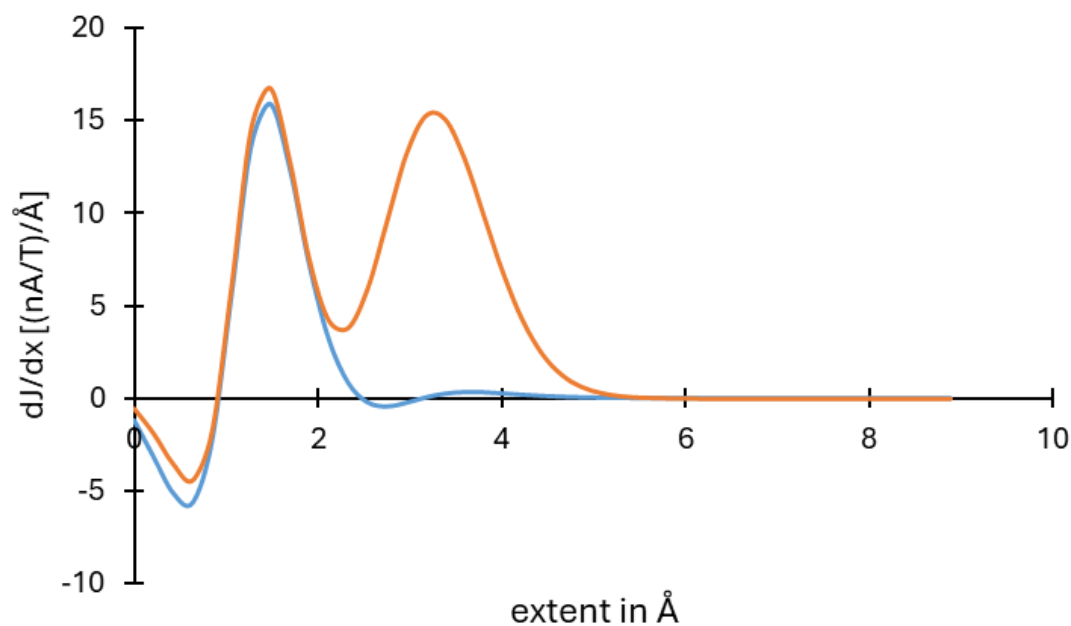
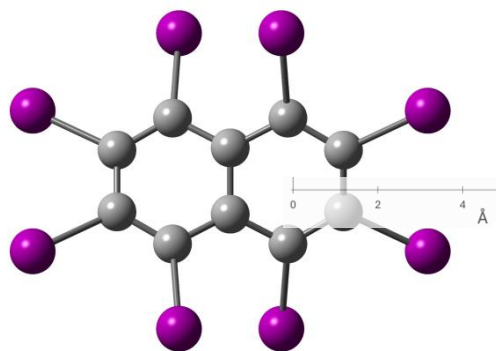


$C_{10}I_8$



$C_{10}I_8^{2+}$

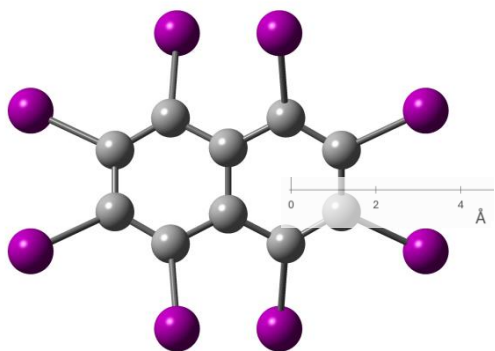
Periodo-naphtalene – bond current strengths profile



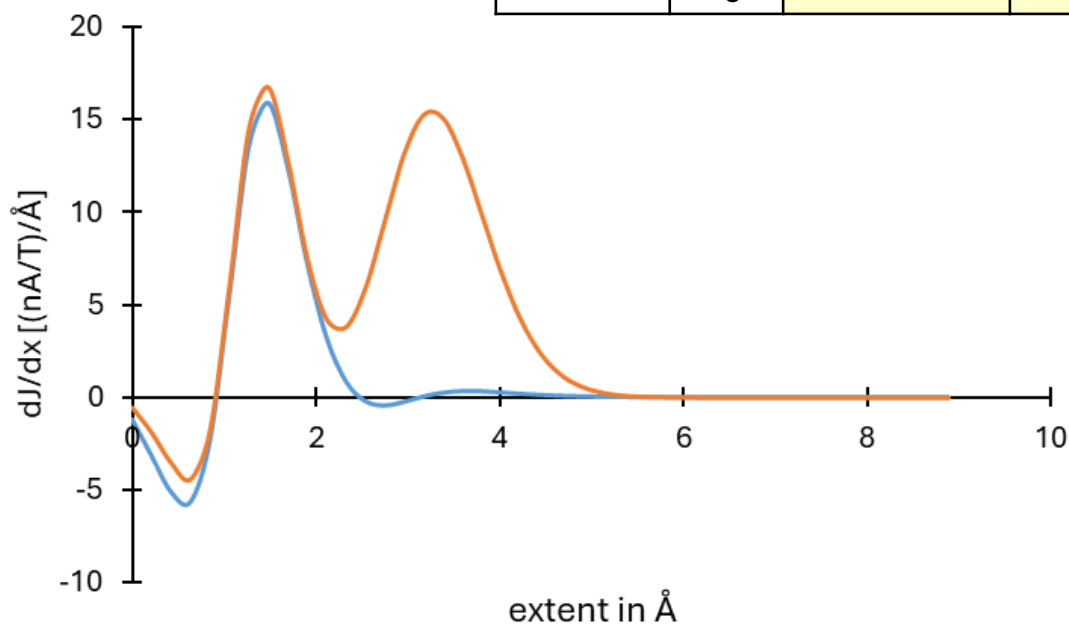
C₁₀I₈ – blue line

C₁₀I₈²⁺ – orange line

Periodo-naphtalene – bond current strengths profile



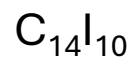
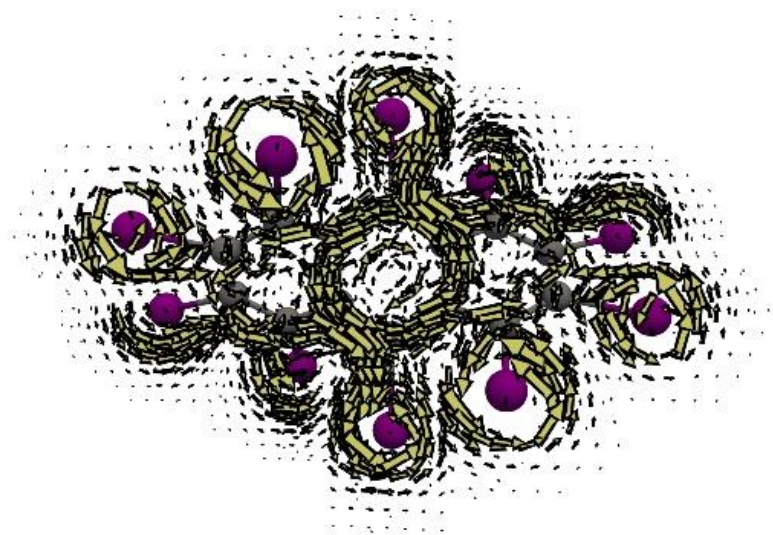
		J [nA T ⁻¹]	EDDB	I_{ring}
$C_{10}I_8$	C_{10}	8.6	5.918	0.573
	I_8	0.3	0.252	0.080
$C_{10}I_8^{2+}$	C_{10}	10.5	7.085	0.590
	I_8	21.4	5.825	0.165



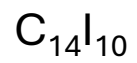
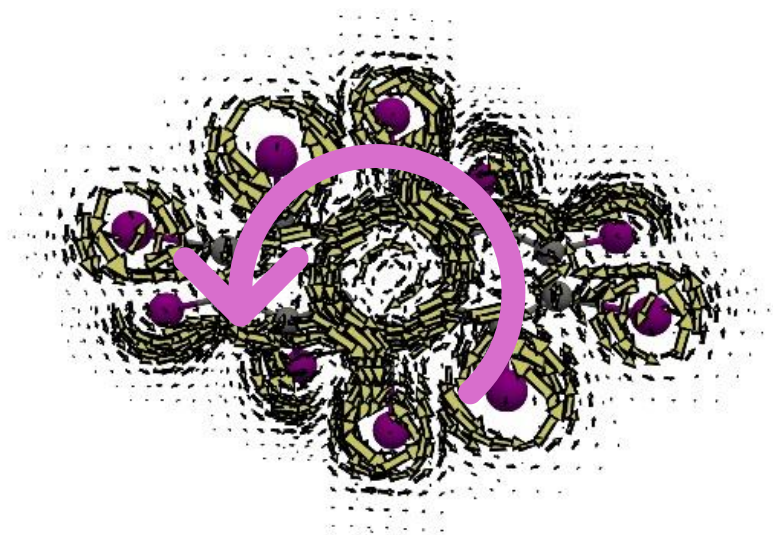
$C_{10}I_8$ – blue line

$C_{10}I_8^{2+}$ – orange line

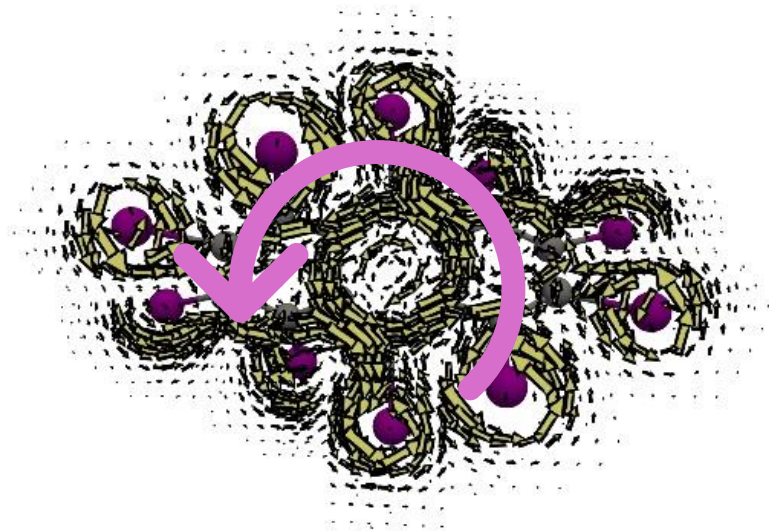
Periodo-anthracene



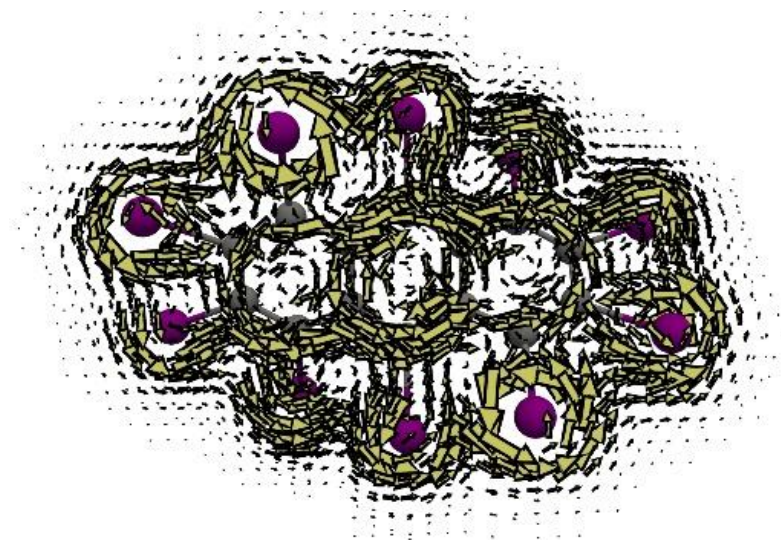
Periodo-anthracene



Periodo-anthracene

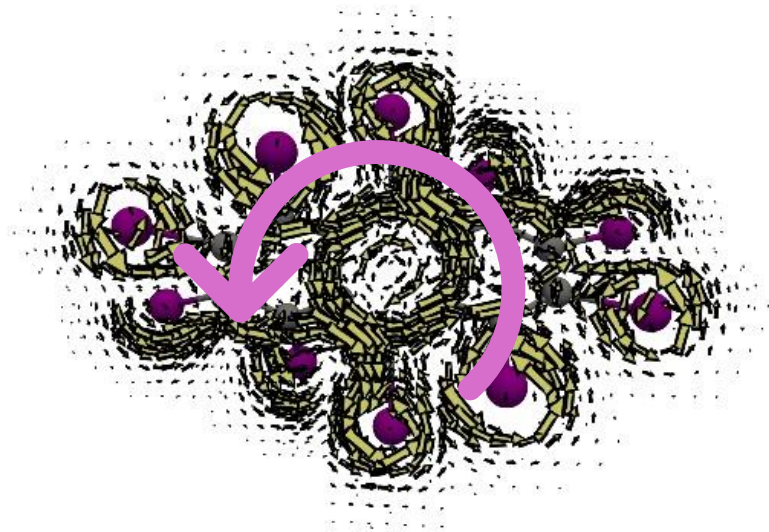


$C_{14}I_{10}$



$C_{14}I_{10}^{2+}$

Periodo-anthracene

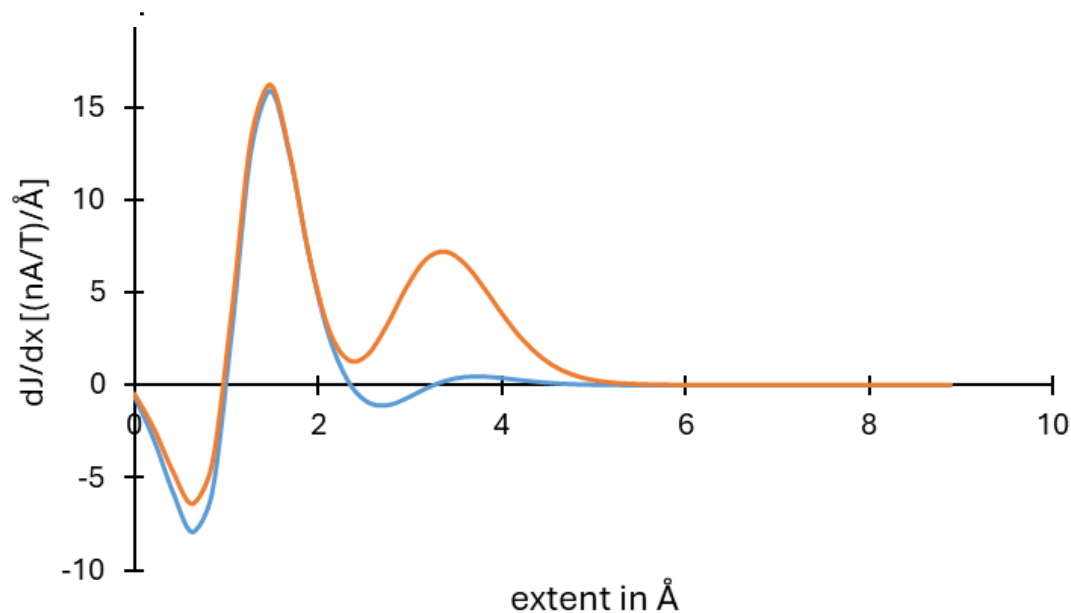
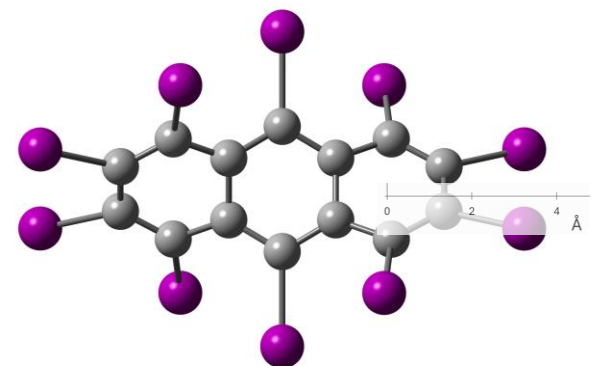


$C_{14}I_{10}$



$C_{14}I_{10}^{2+}$

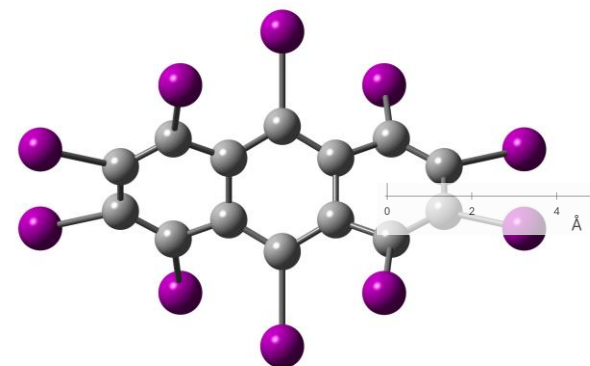
Periodo-anthracene – bond current strengths profile



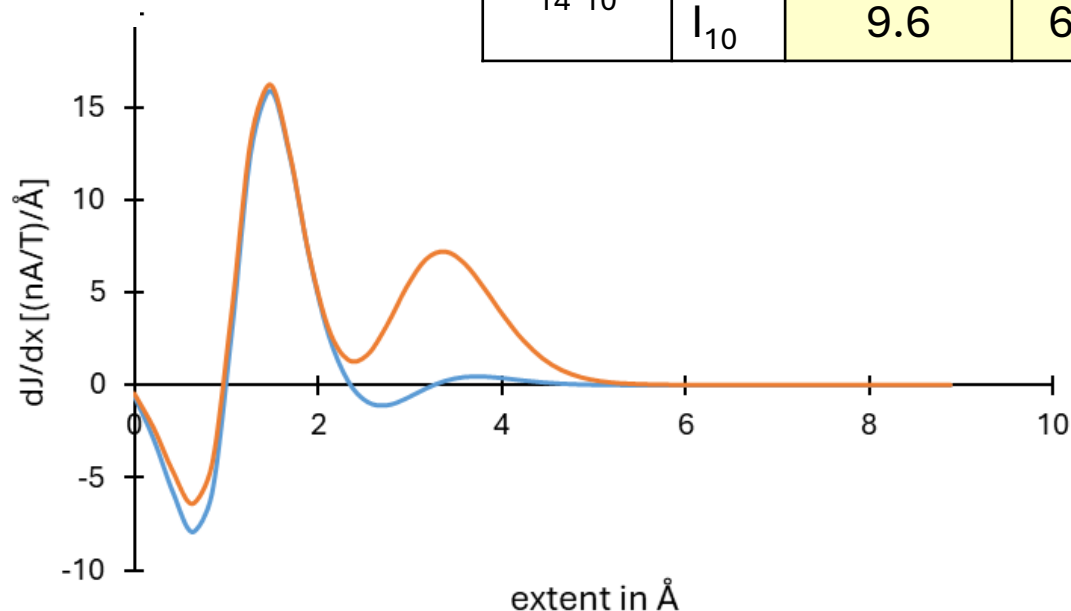
$C_{14}I_{10}$ – blue line

$C_{14}I_{10}^{2+}$ – orange line

Periodo-anthracene – bond current strengths profile



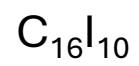
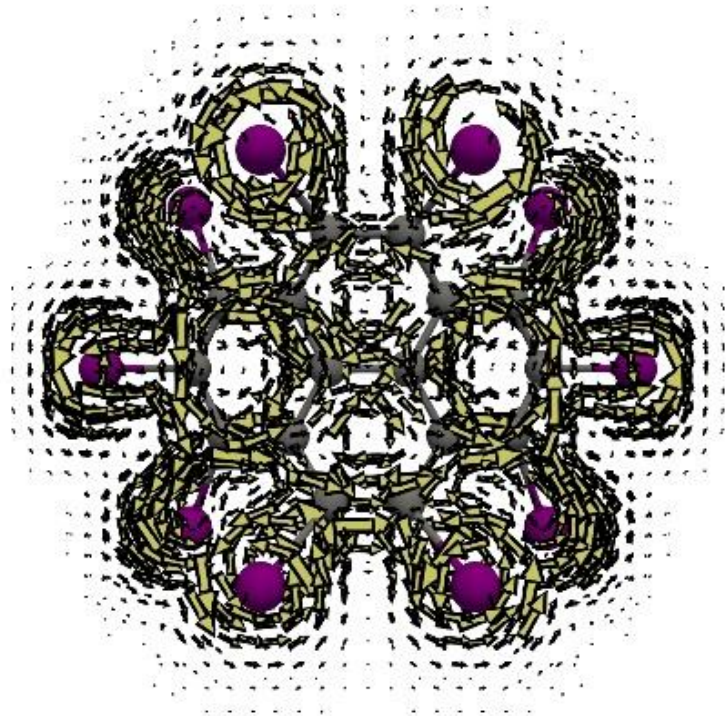
		$J [\text{nA T}^{-1}]$	EDDB	I_{ring}
$C_{14}I_{10}$	C_{14}	6.4	8.130	0.556
	I_{10}	-0.2	0.301	0.066
$C_{14}I_{10}^{2+}$	C_{14}	8.3	8.916	0.570
	I_{10}	9.6	6.089	0.165



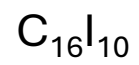
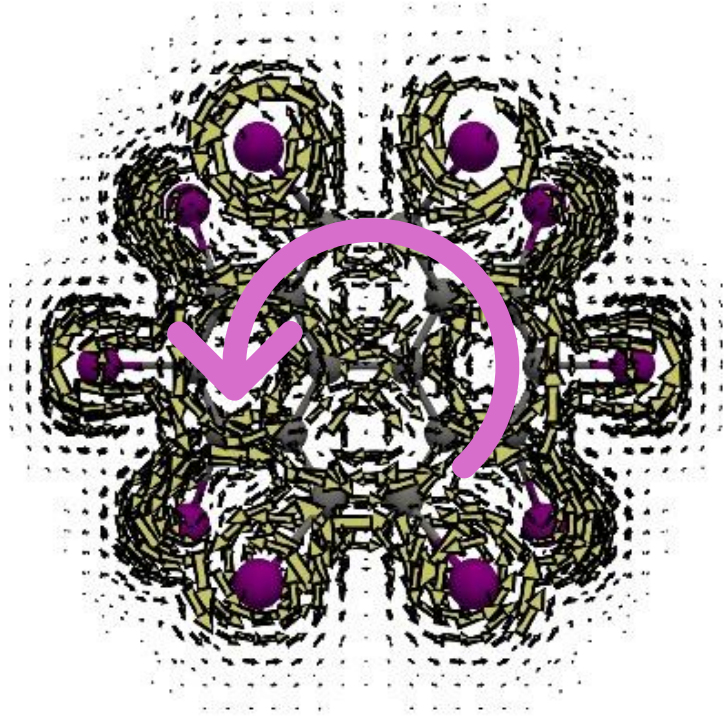
$C_{14}I_{10}$ – blue line

$C_{14}I_{10}^{2+}$ – orange line

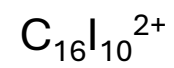
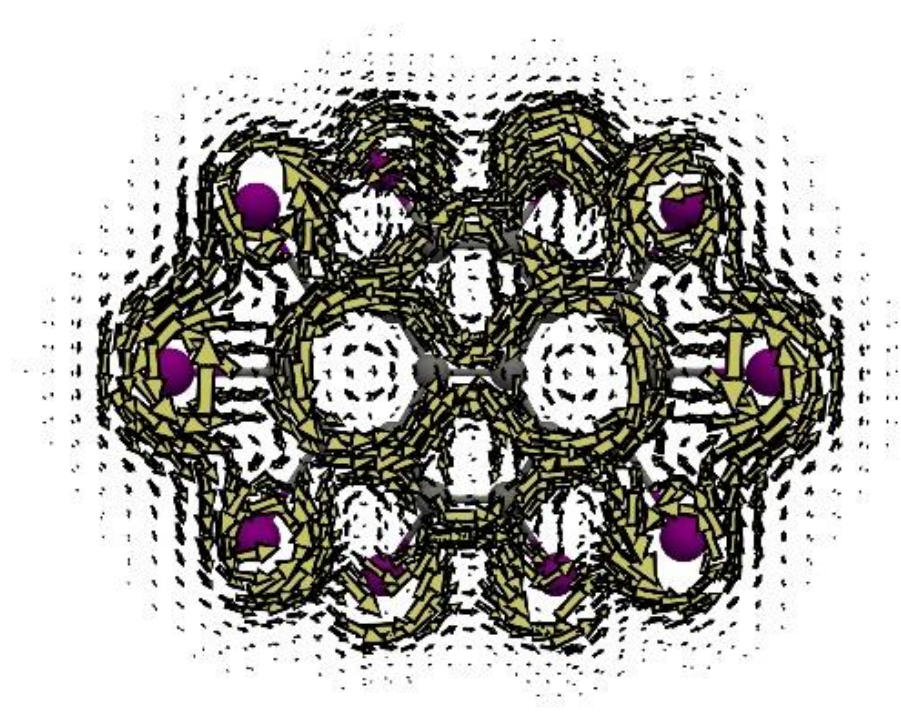
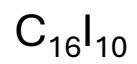
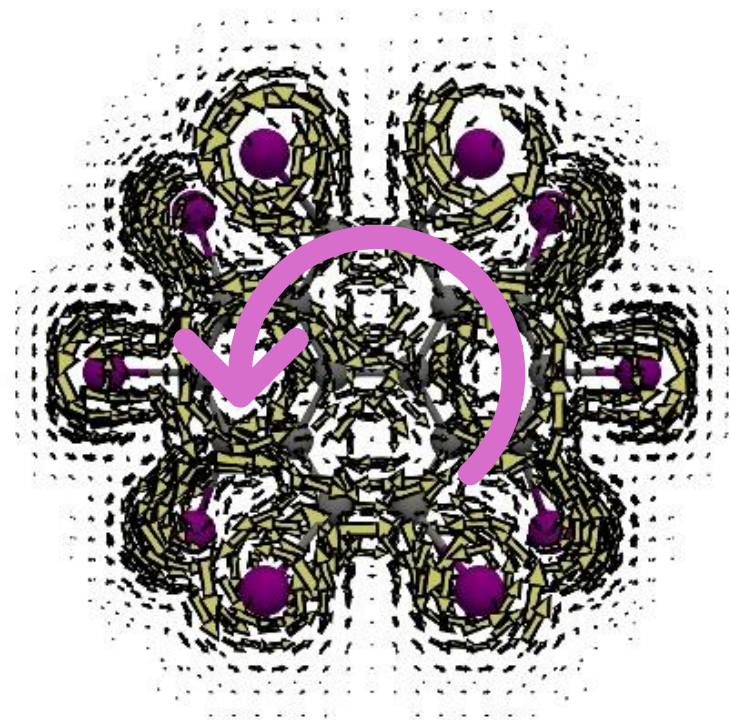
Periodo-perylene



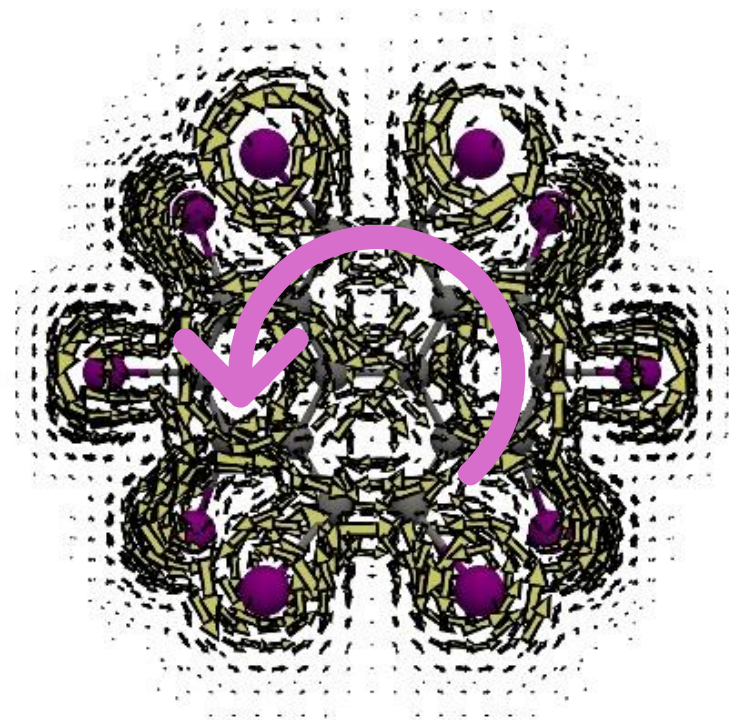
Periodo-perylene



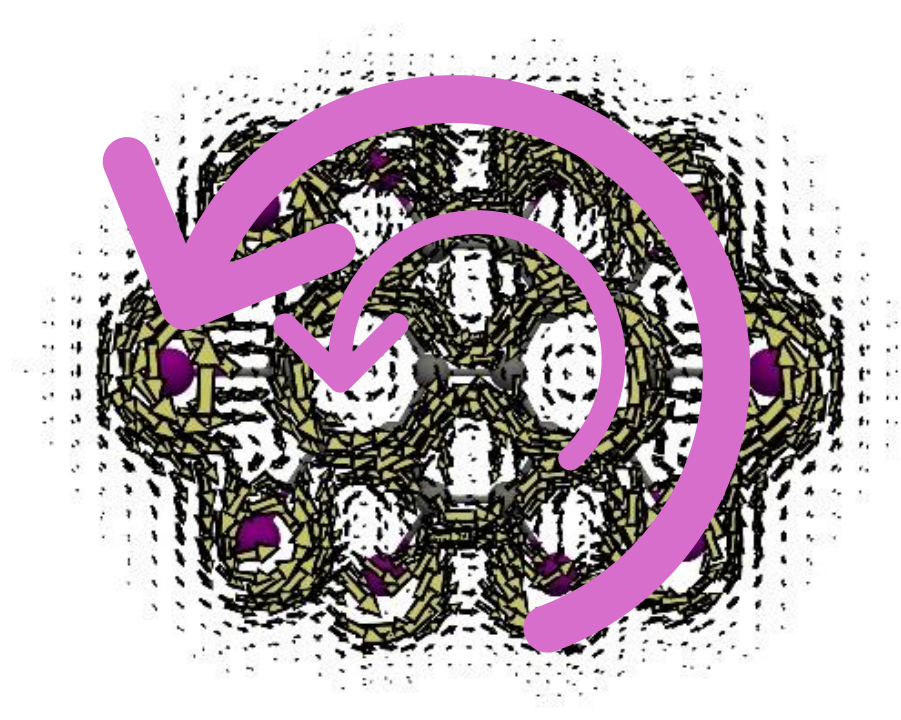
Periodo-perylene



Periodo-perylene

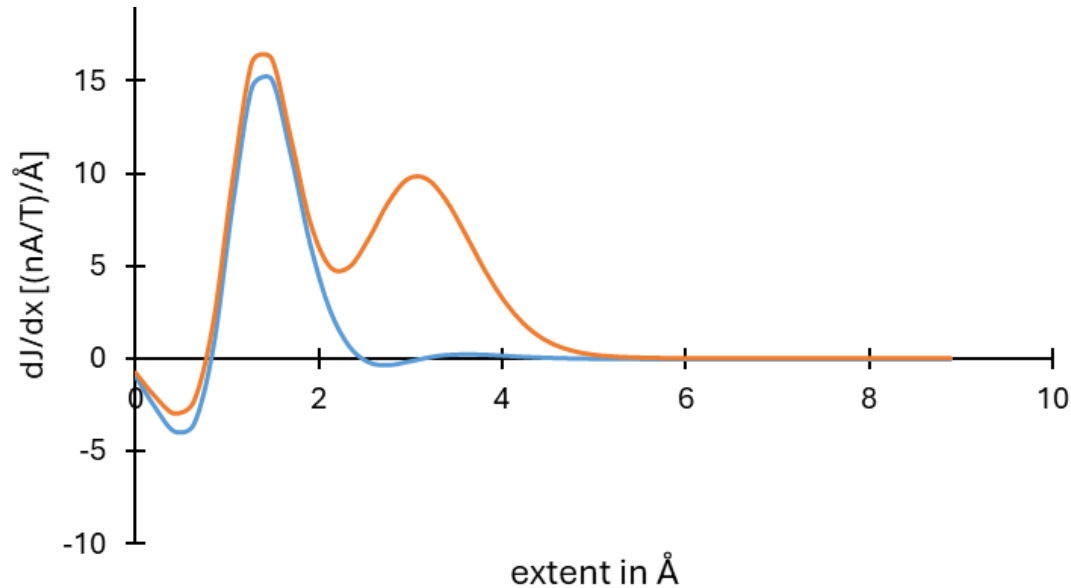
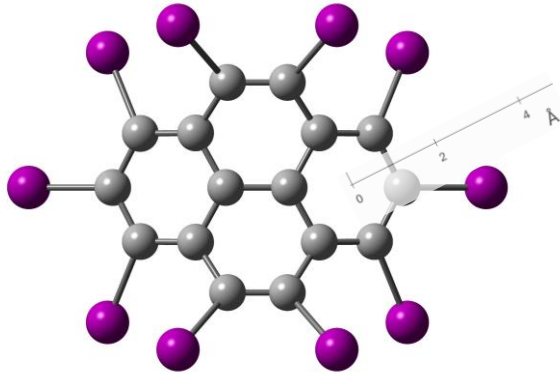


$C_{16}I_{10}$



$C_{16}I_{10}^{2+}$

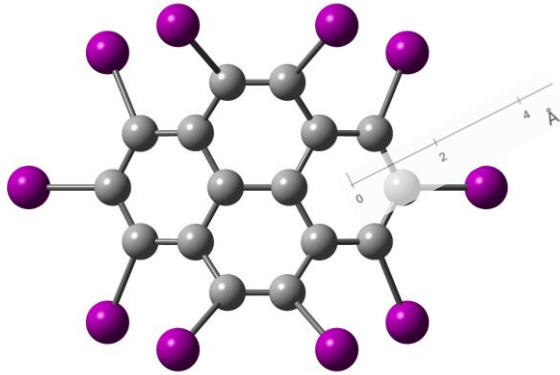
Periodo-perylene – bond current strengths profile



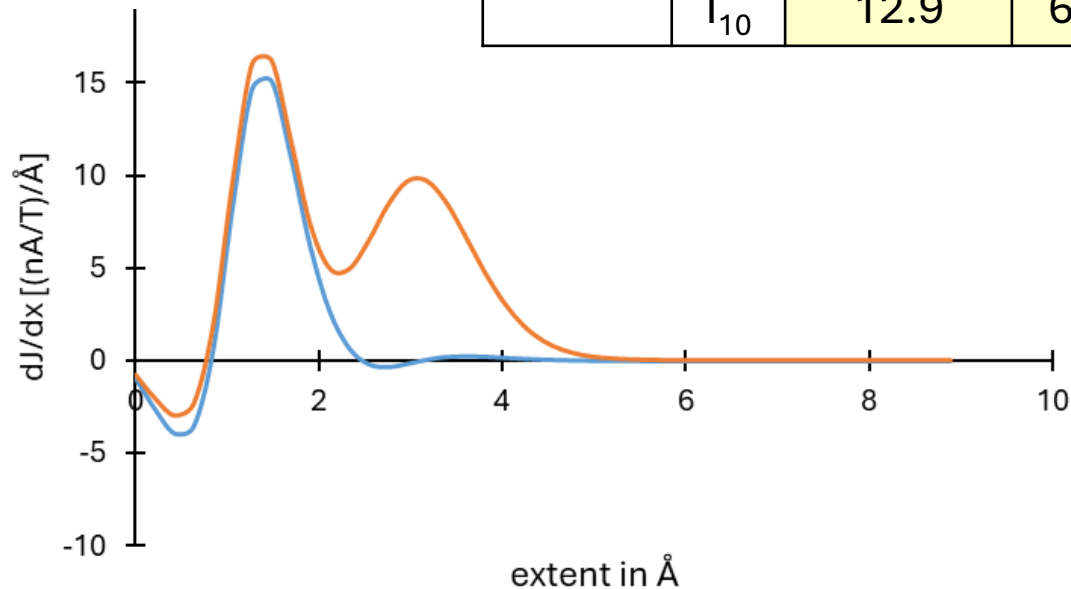
$C_{16}I_{10}$ – blue line

$C_{16}I_{10}^{2+}$ – orange line

Periodo-perylene – bond current strengths profile



		$J [\text{nA T}^{-1}]$	EDDB	I_{ring}
$\text{C}_{16}\text{I}_{10}$	C_{16}	10.2	7.233	0.542
	I_{10}	0.1	0.266	0.087
$\text{C}_{16}\text{I}_{10}^{2+}$	C_{16}	13.7	7.656	0.560
	I_{10}	12.9	6.146	0.172



$\text{C}_{16}\text{I}_{10}$ – blue line

$\text{C}_{16}\text{I}_{10}^{2+}$ - orange line

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- It was found that bond strengths profiles are useful tool in double aromaticity studies.
- Our results indicate that *periodo*-naphthalene dication, *periodo*-anthracene dication and *periodo*-perylene dication are **double aromatic** systems.
- On the other side, *periodo*-pentalene dication is π -aromatic system.